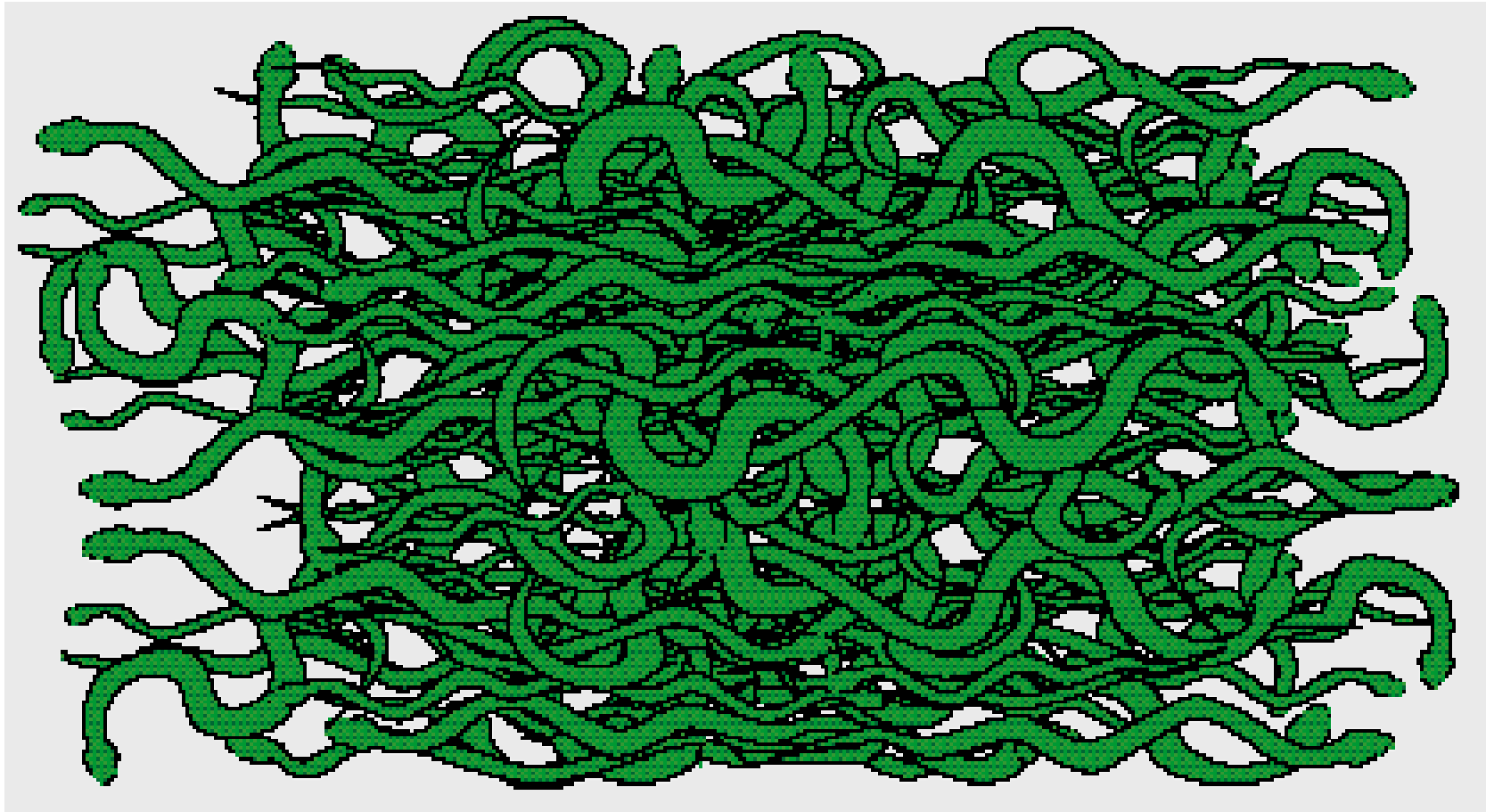


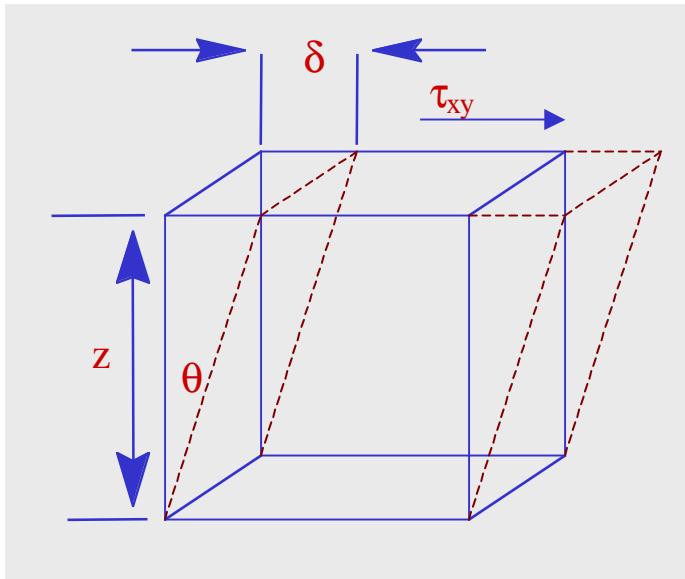
Topics to be Covered

- *Polymer Melt Rheology*
- *Introduction to Viscoelastic Behavior*
- *Time-Temperature Equivalence*

Chapter 11 in CD (Polymer Science and Engineering)

Polymer Melt Rheology





Newton's Law

The most convenient way to describe deformation under shear is in terms of the angle θ through which the material is deformed;

$$\tan \theta = \frac{\delta}{z} = \gamma_{xy}$$

Then if you shear a fluid at a constant rate

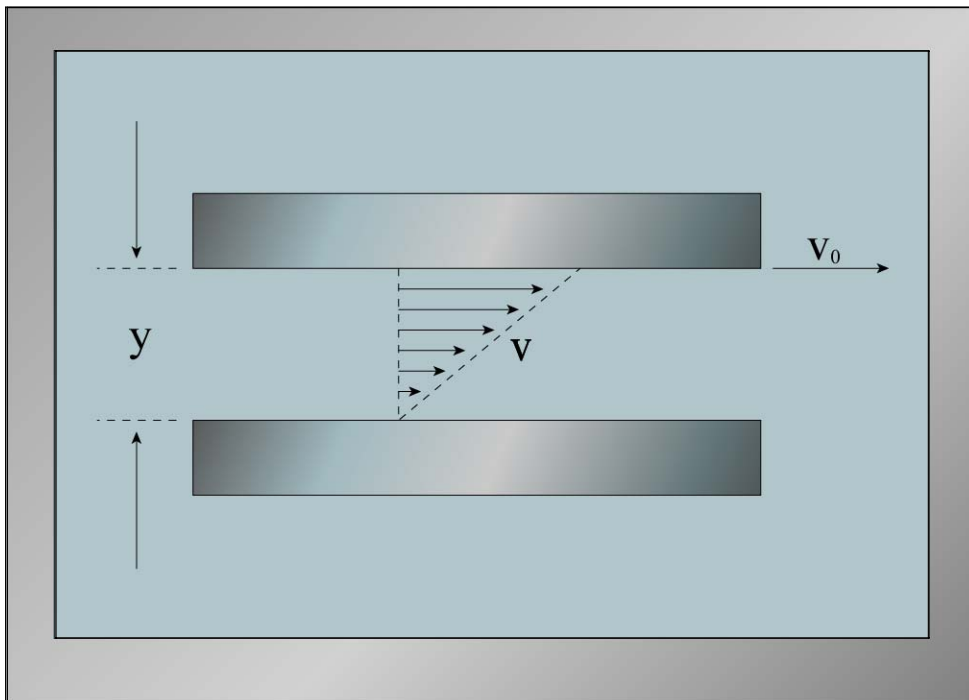
$$\dot{\gamma}_{xy} = \frac{\dot{\delta}}{z} = \frac{v_0}{z}$$

$$\dot{\gamma}_{xy} = \frac{dv}{dz} = \frac{d}{dt} \left[\frac{dx}{dz} \right]$$

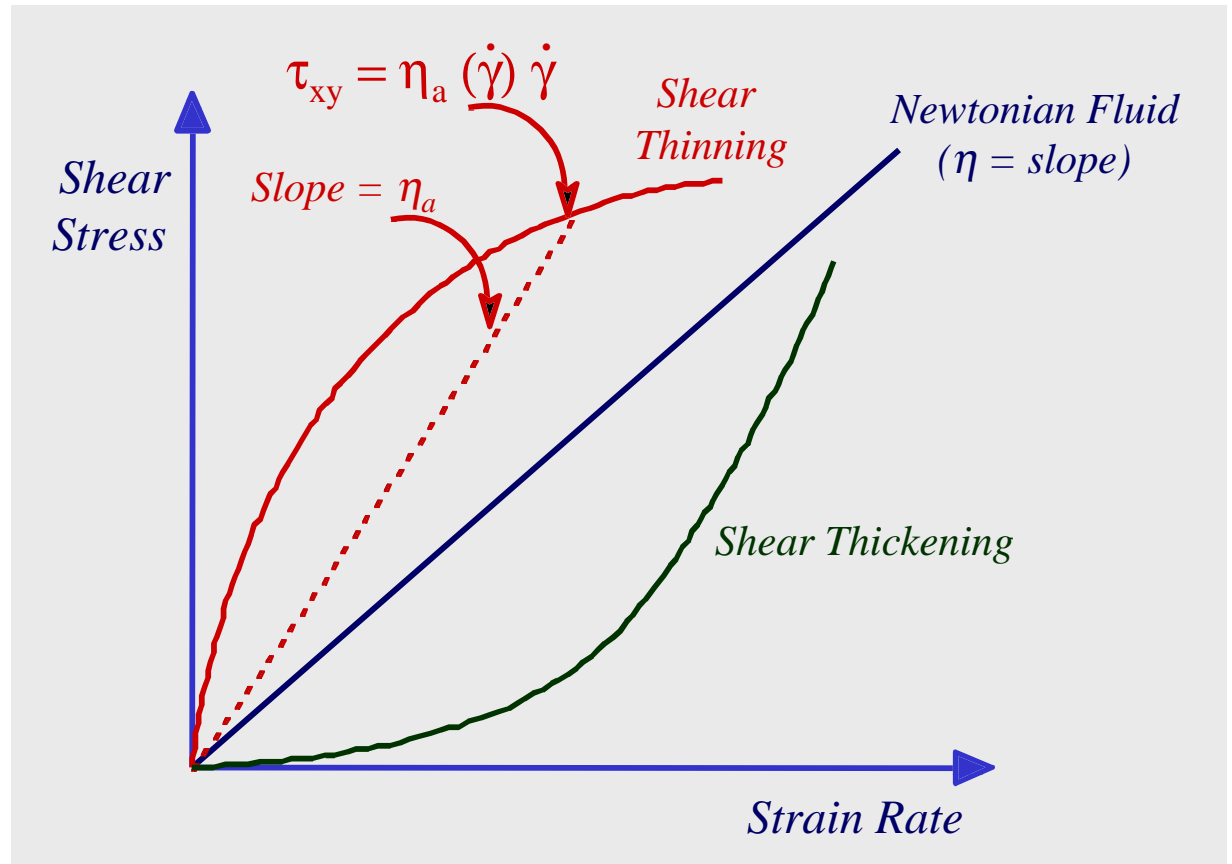
Just as stress is proportional to strain in Hooke's law for a solid, the shear stress is proportional to the rate of strain for a fluid

$$\tau_{xy} \propto \dot{\gamma}_{xy}$$

$$\tau_{xy} = \eta \dot{\gamma}_{xy}$$

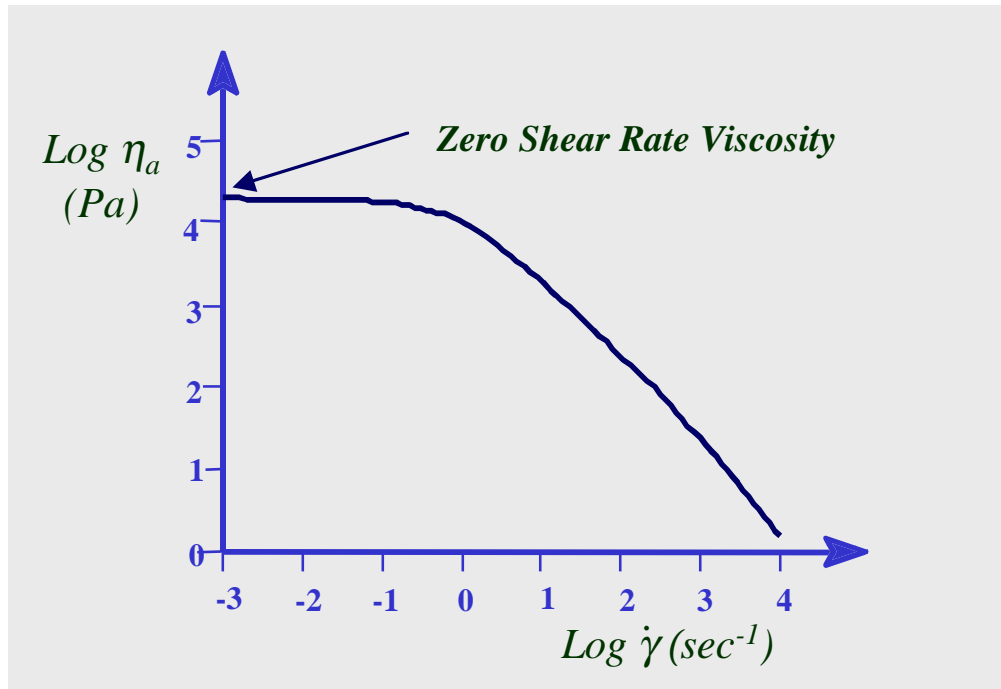


Newtonian and Non - Newtonian Fluids

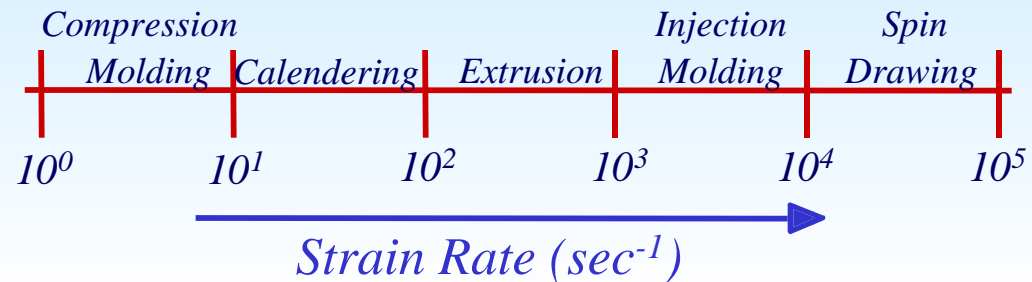


When the chips are down, what do you use?

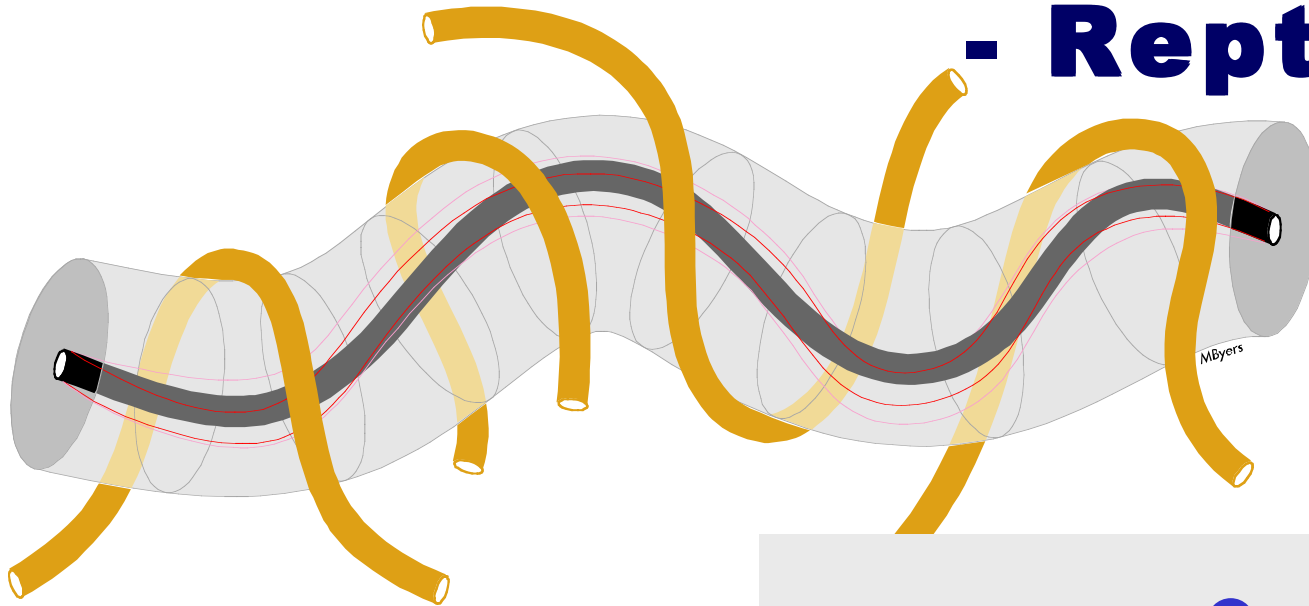
Variation of Melt Viscosity with Strain Rate



SHEAR RATES ENCOUNTERED IN PROCESSING

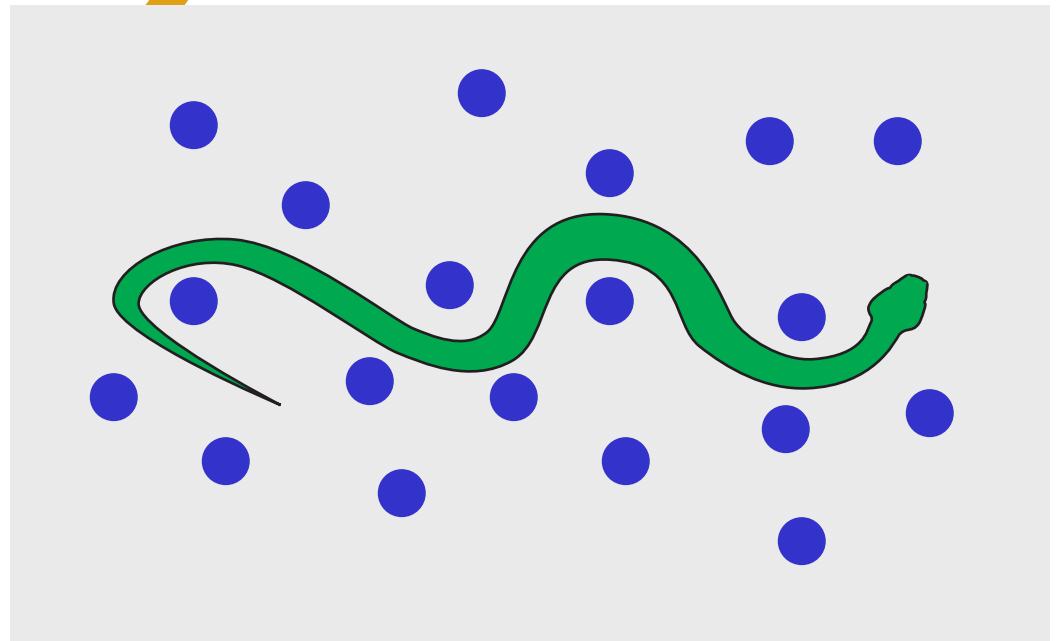


How Do Chains Move - Reptation

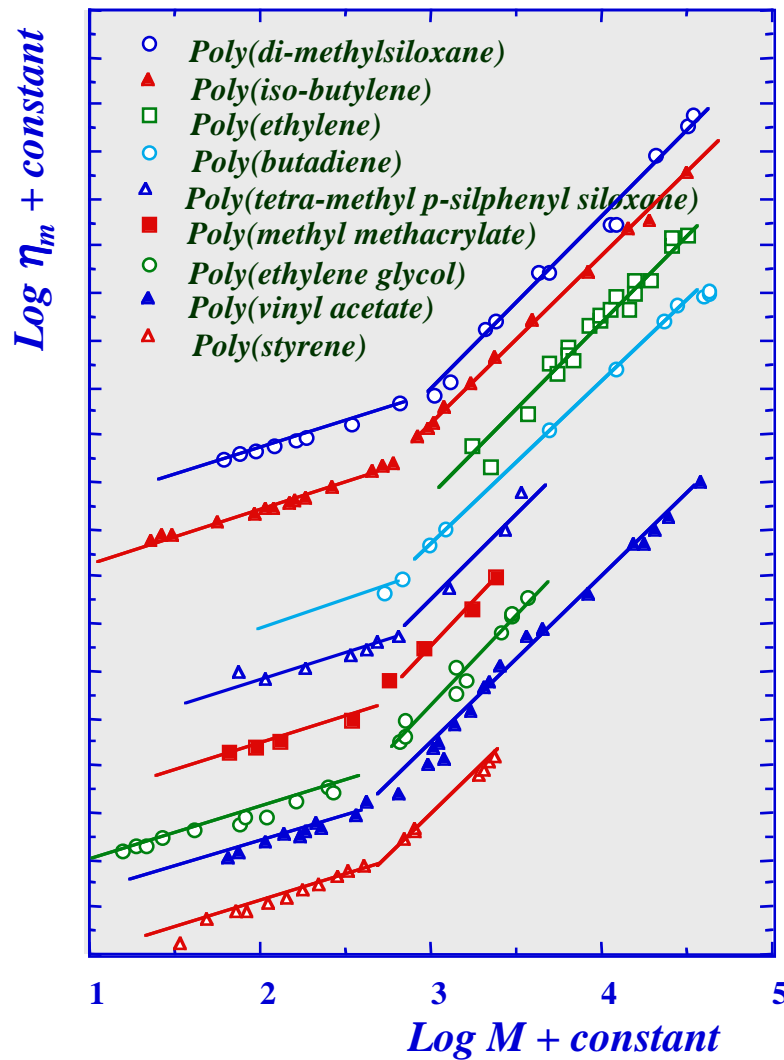


$$D \sim 1/M^2$$

$$\eta_0 \sim M^3$$



Variation of Melt Viscosity with Molecular Weight



$$\eta_m = K_L(DP)^{1.0}$$

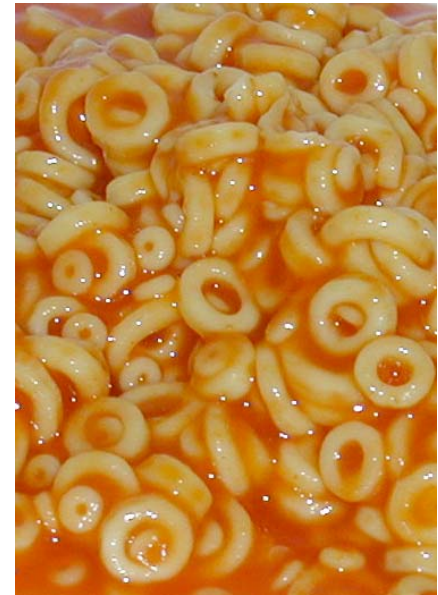
$$\eta_m = K_H(DP)^{3.4}$$

Redrawn from the data of G. C. Berry and T. G. Fox, *Adv. Polym. Sci.*, 5, 261 (1968)

Entanglements



Short chains don't entangle but long ones do - think of the difference between a nice linguini and spaghettios, the little round things you can get out of a tin (we have some value judgements concerning the relative merits of these two forms of pasta, but on the advice of our lawyers we shall refrain from comment).



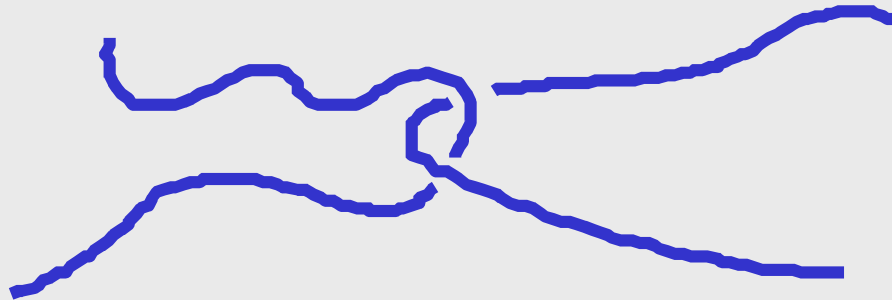
Entanglements

Viscosity - a measure of the frictional forces acting on a molecule

$$\eta_m = K_L(DP)^{1.0}$$

Small molecules - the viscosity varies directly with size

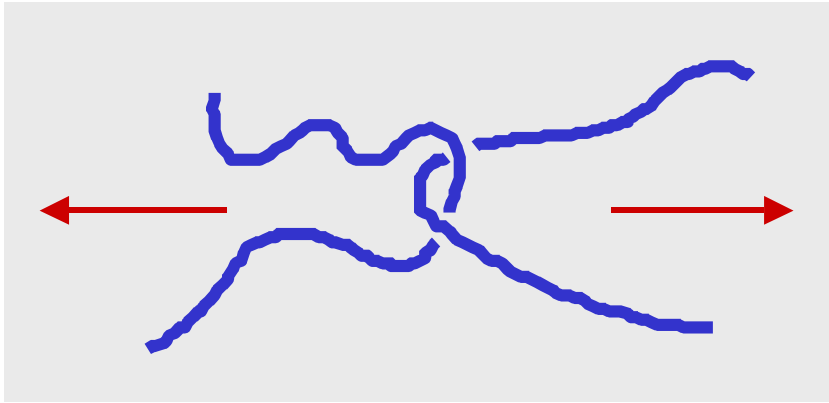
At a critical chain length chains start to become tangled up with one another, however



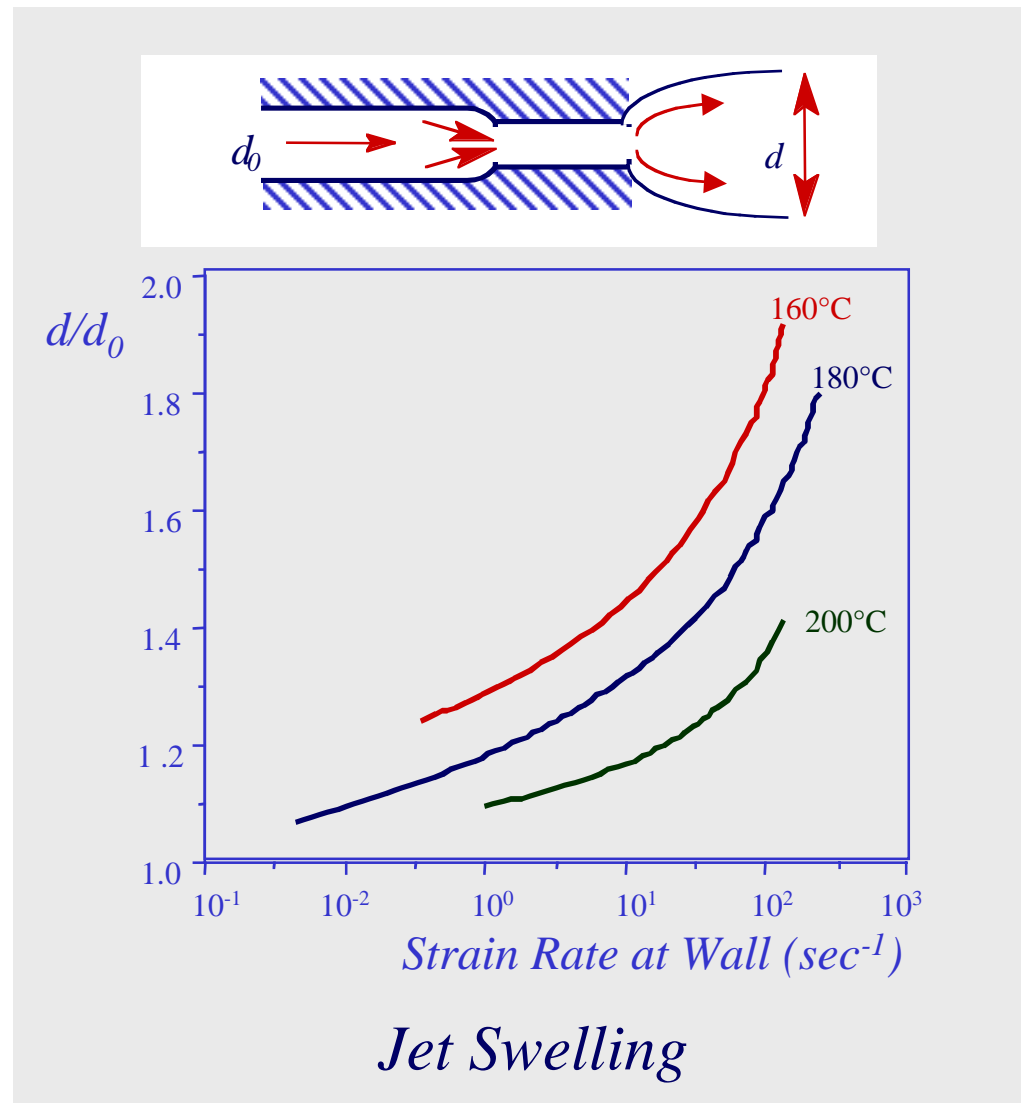
Then

$$\eta_m = K_H(DP)^{3.4}$$

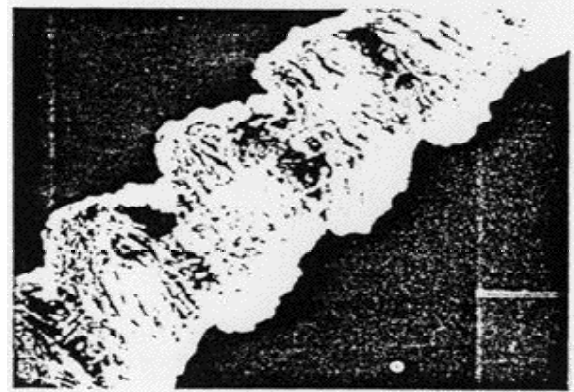
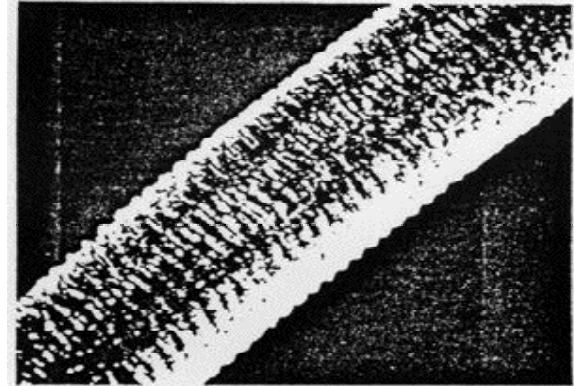
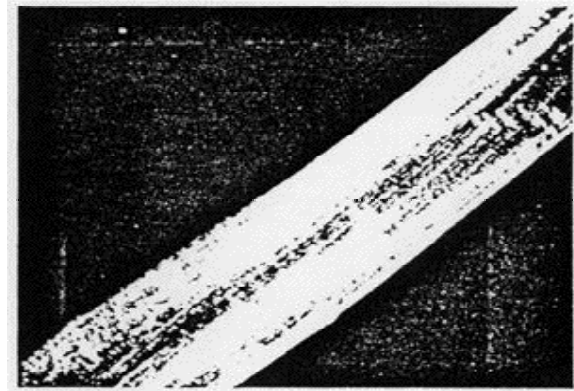
Entanglements and the Elastic Properties of Polymer Melts



Depending upon the rate at which chains disentangle relative to the rate at which they stretch out, there is an elastic component to the behavior of polymer melts. There are various consequences as a result of this.



Melt Fracture



Reproduced with permission from J. J. Benbow,
R. N. Browne and E. R. Howells, *Coll. Intern.
Rheol., Paris*, June-July 1960.

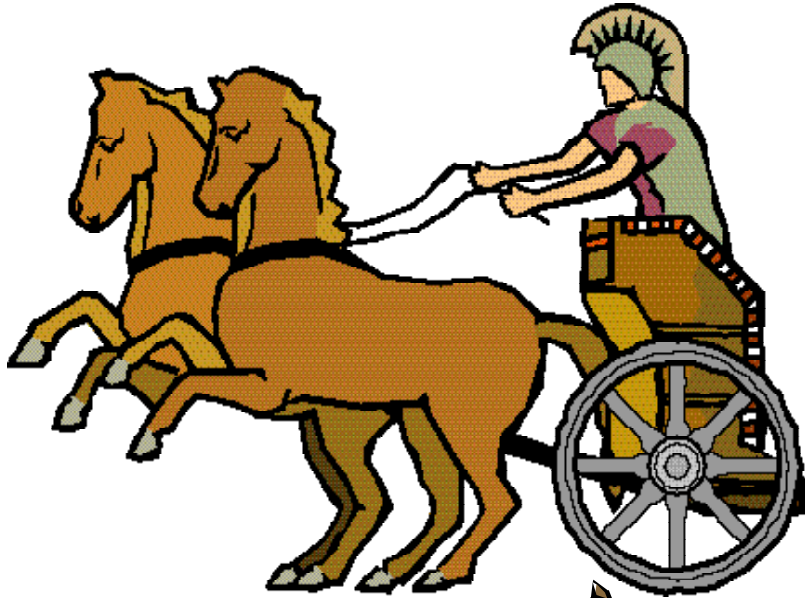
Viscoelasticity

If we stretch a crystalline solid,
The energy is stored in the
Chemical bonds

If we apply a shear stress to
A fluid, energy is dissipated
In flow

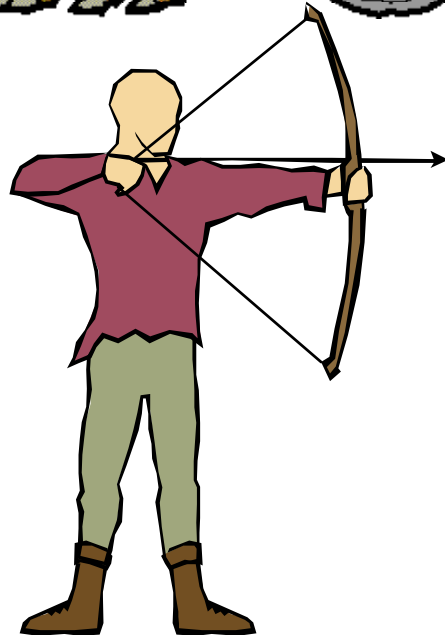


Viscoelasticity



Homer knew that the first thing To do on getting your chariot out in the morning was to put the wheels back in.

(Telemachus, in *The Odyssey*, would tip his chariot against a wall)

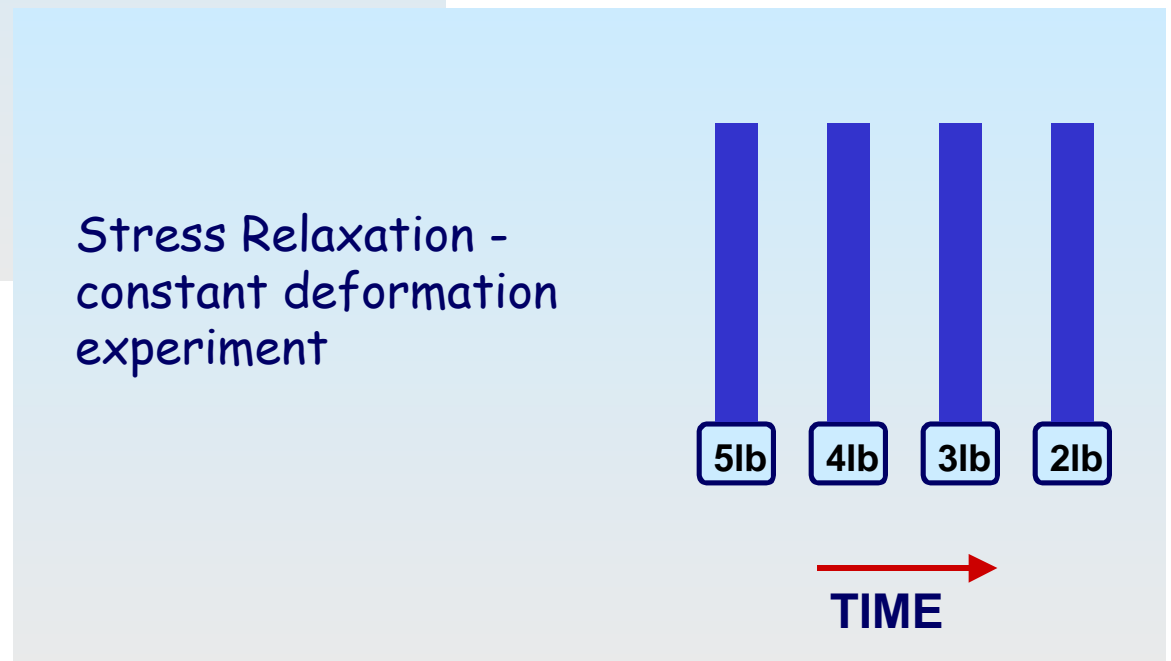
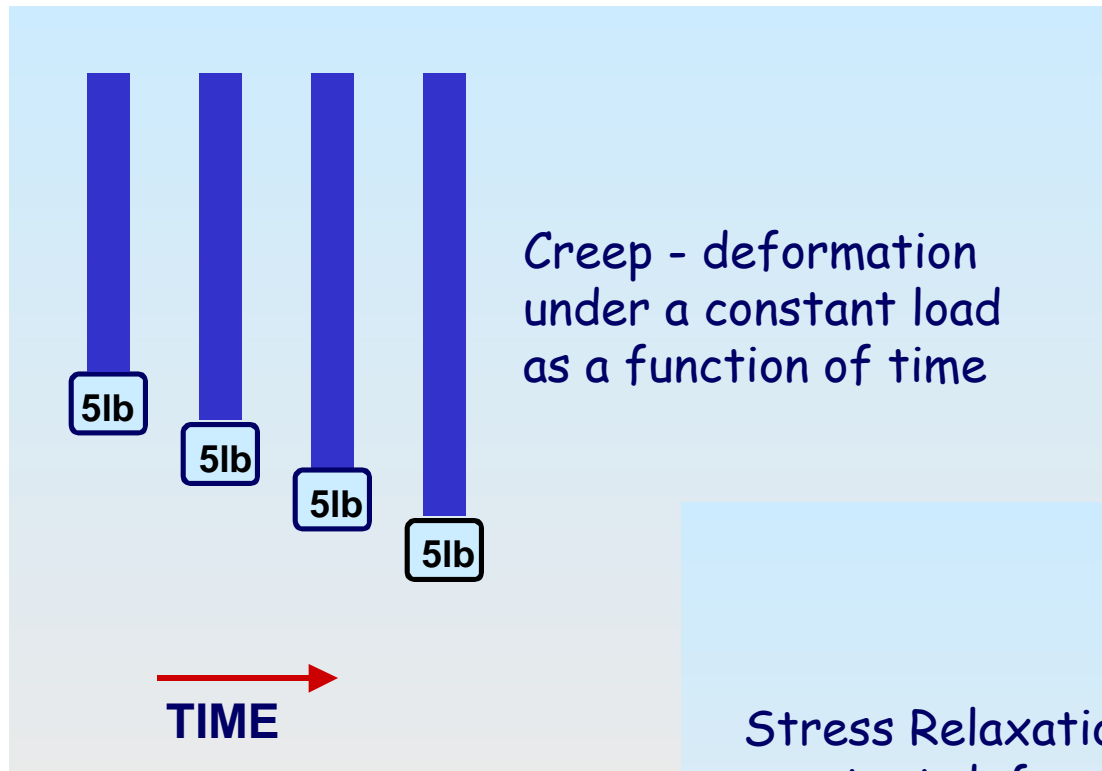


Robin hood knew never to leave his bow strung

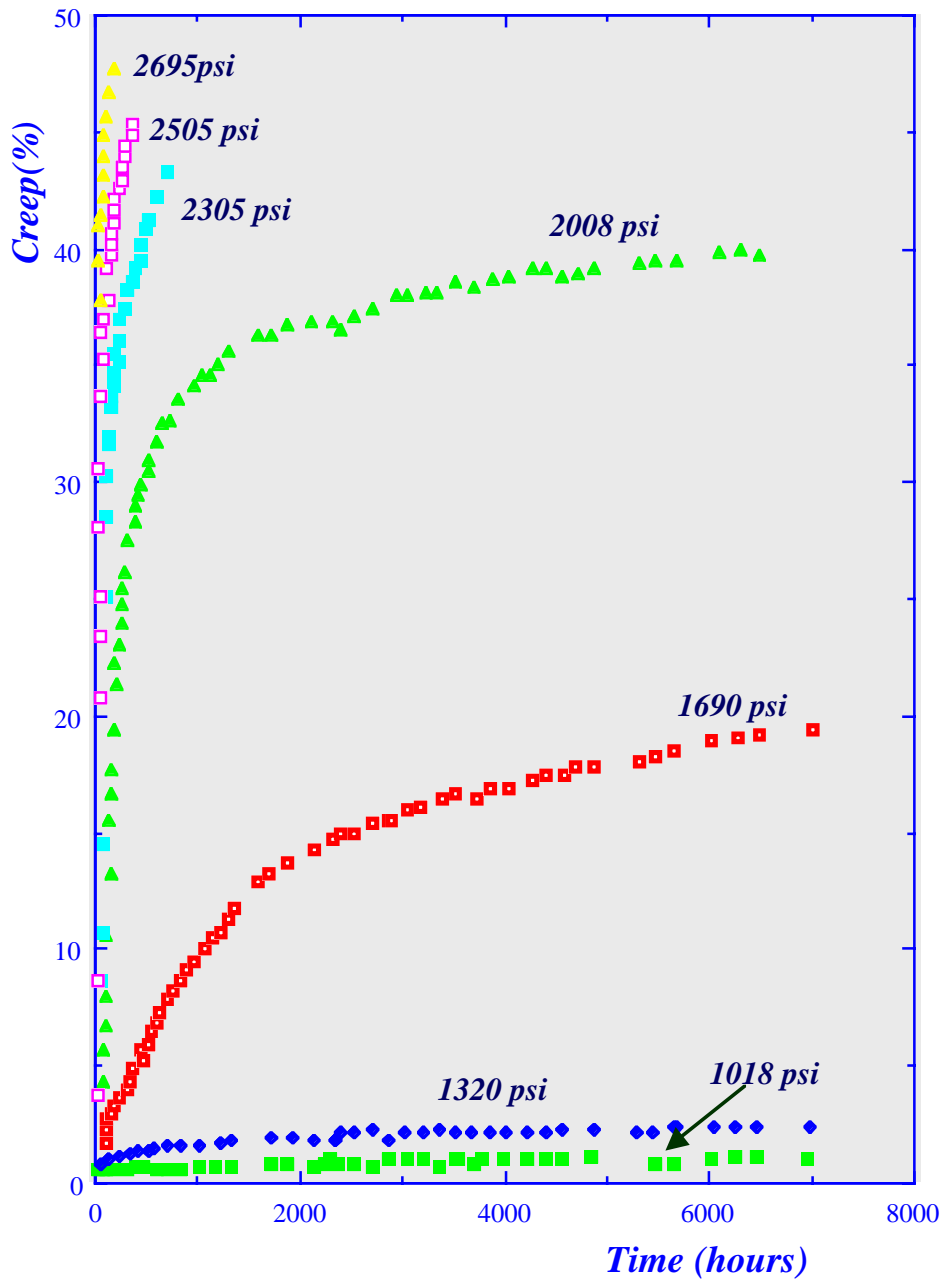
Experimental Observations

- *Creep*
- *Stress Relaxation*
- *Dynamic Mechanical Analysis*

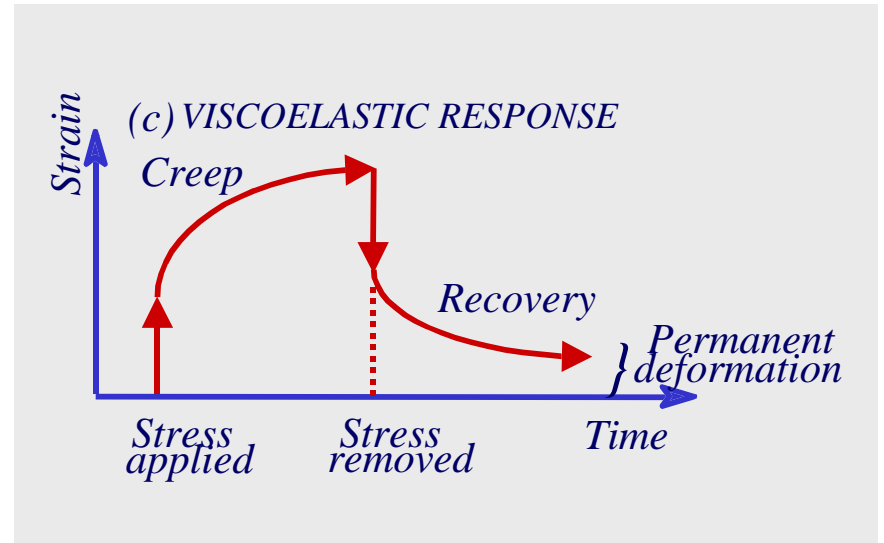
Creep and Stress Relaxation



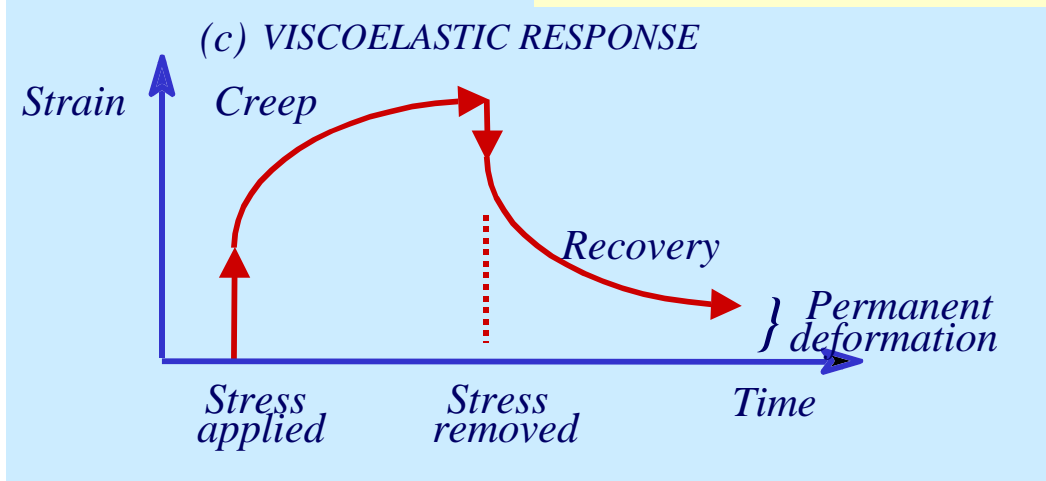
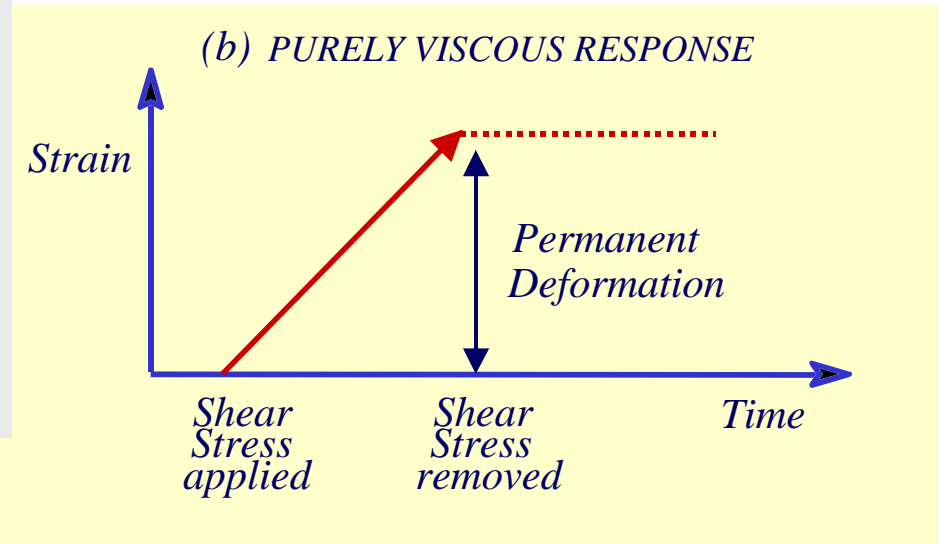
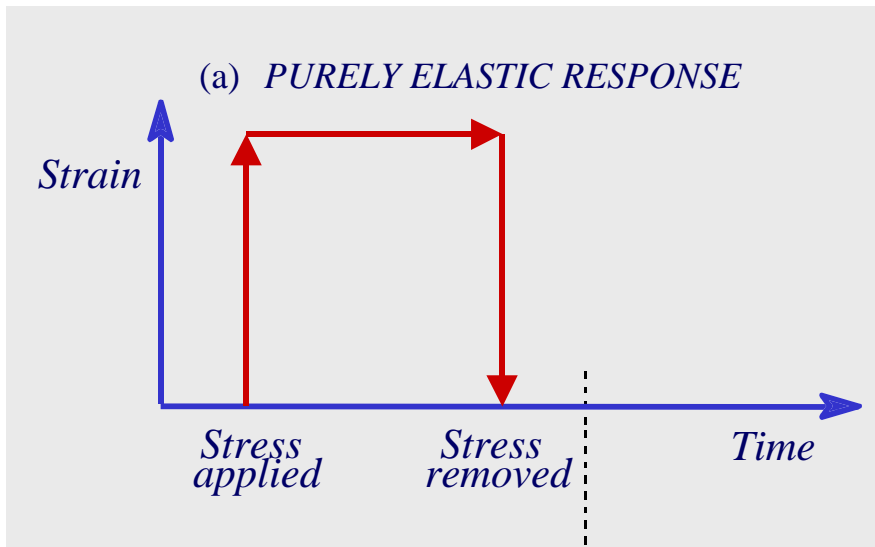
Redrawn from the data of W. N. Findley, *Modern Plastics*, 19, 71 (August 1942)



Creep and Recovery



Strain vs. Time Plots - Creep

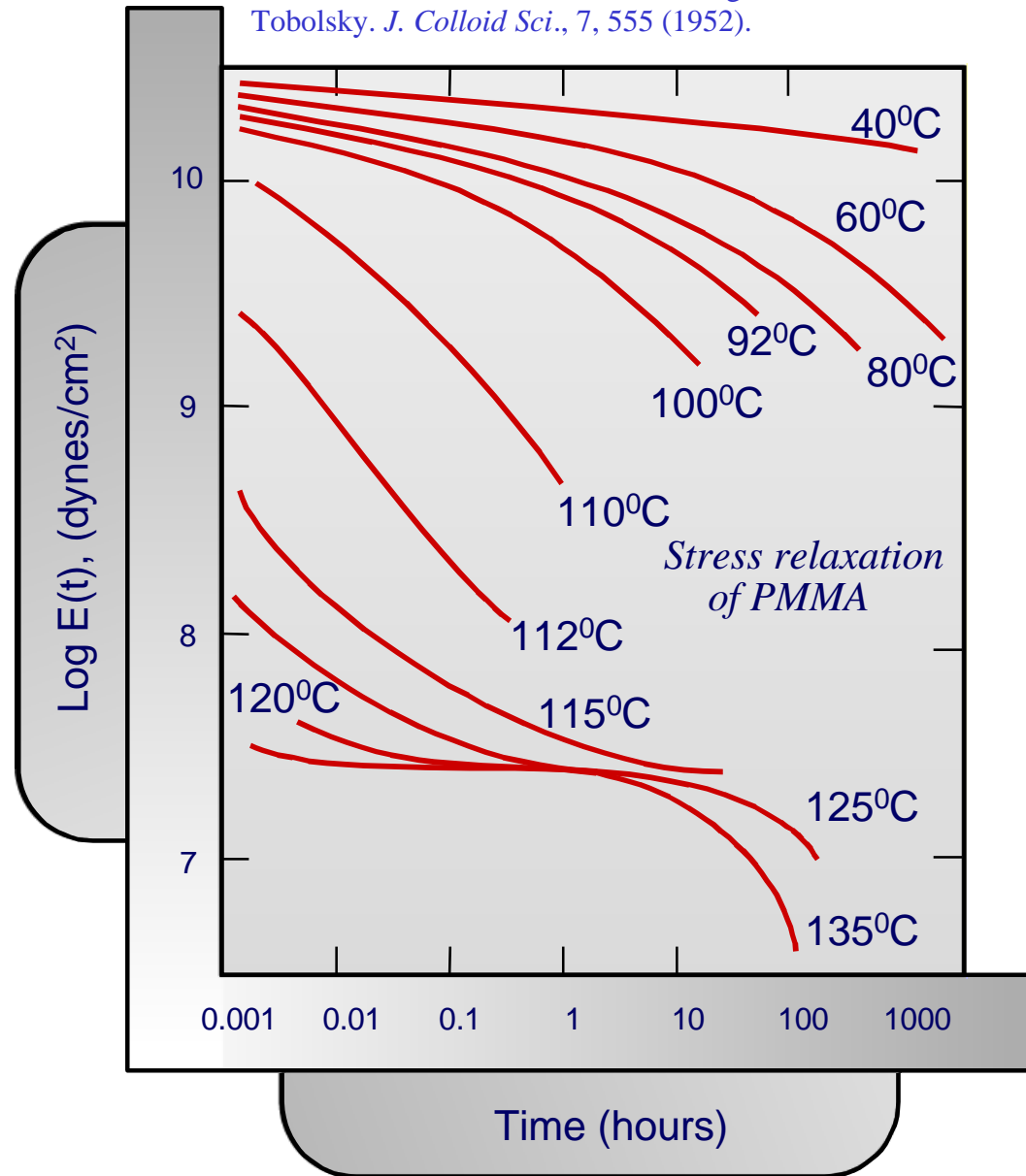


Stress Relaxation

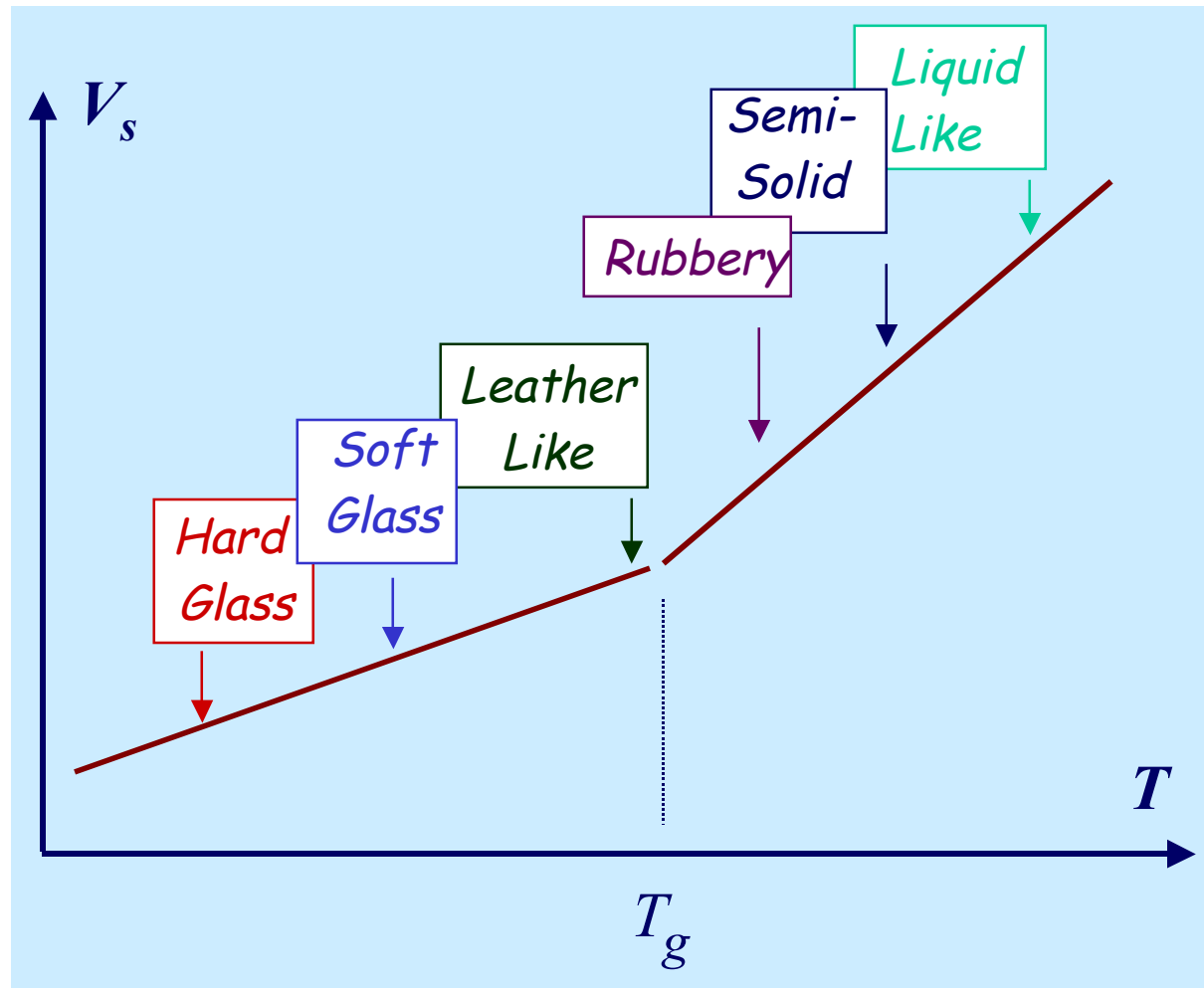
The data are not usually reported as a stress/time plot, but as a modulus/time plot. This time dependent modulus, called the *relaxation modulus*, is simply the time dependent stress divided by the (constant) strain

$$E(t) = \frac{\sigma(t)}{\epsilon_0}$$

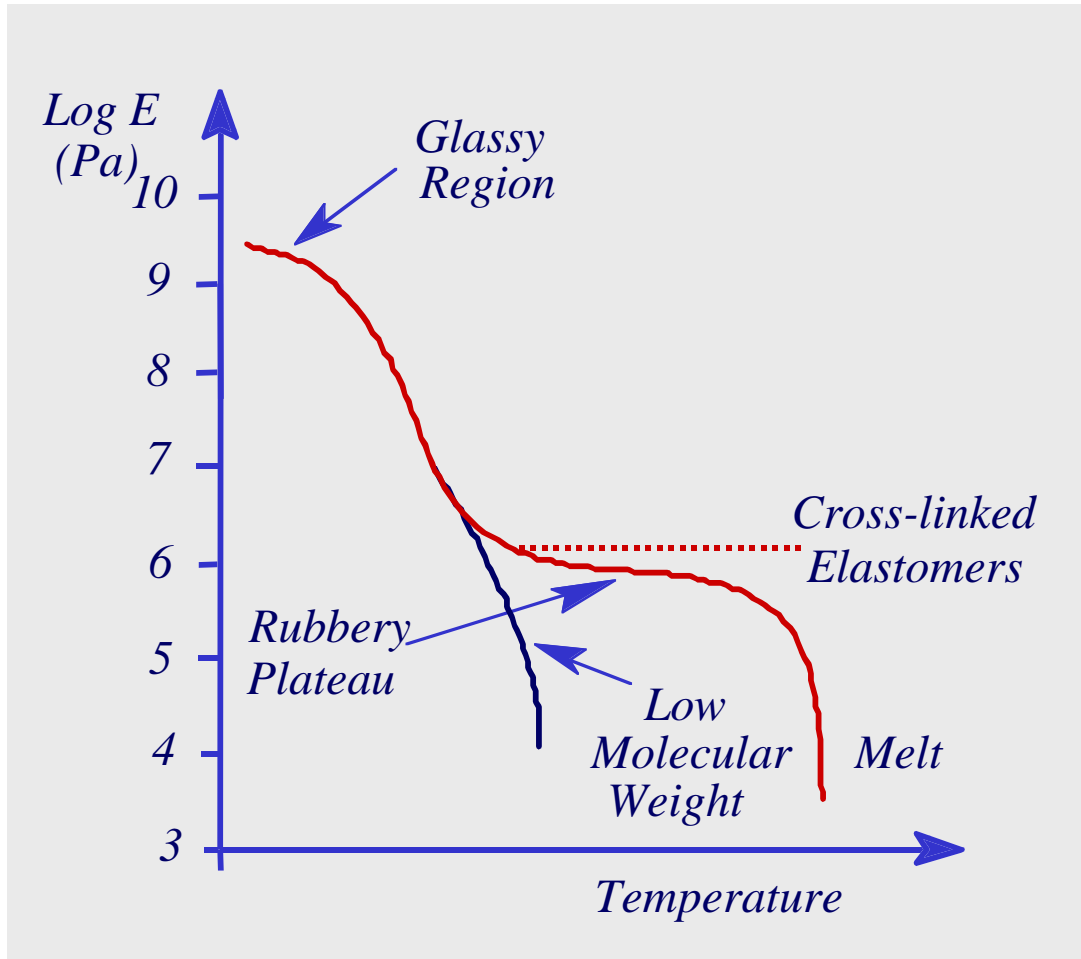
Redrawn from the data of J.R. McLoughlin and A.V. Tobolsky. *J. Colloid Sci.*, 7, 555 (1952).



Amorphous Polymers - Range of Viscoelastic Behaviour



Viscoelastic Properties Of Amorphous Polymers

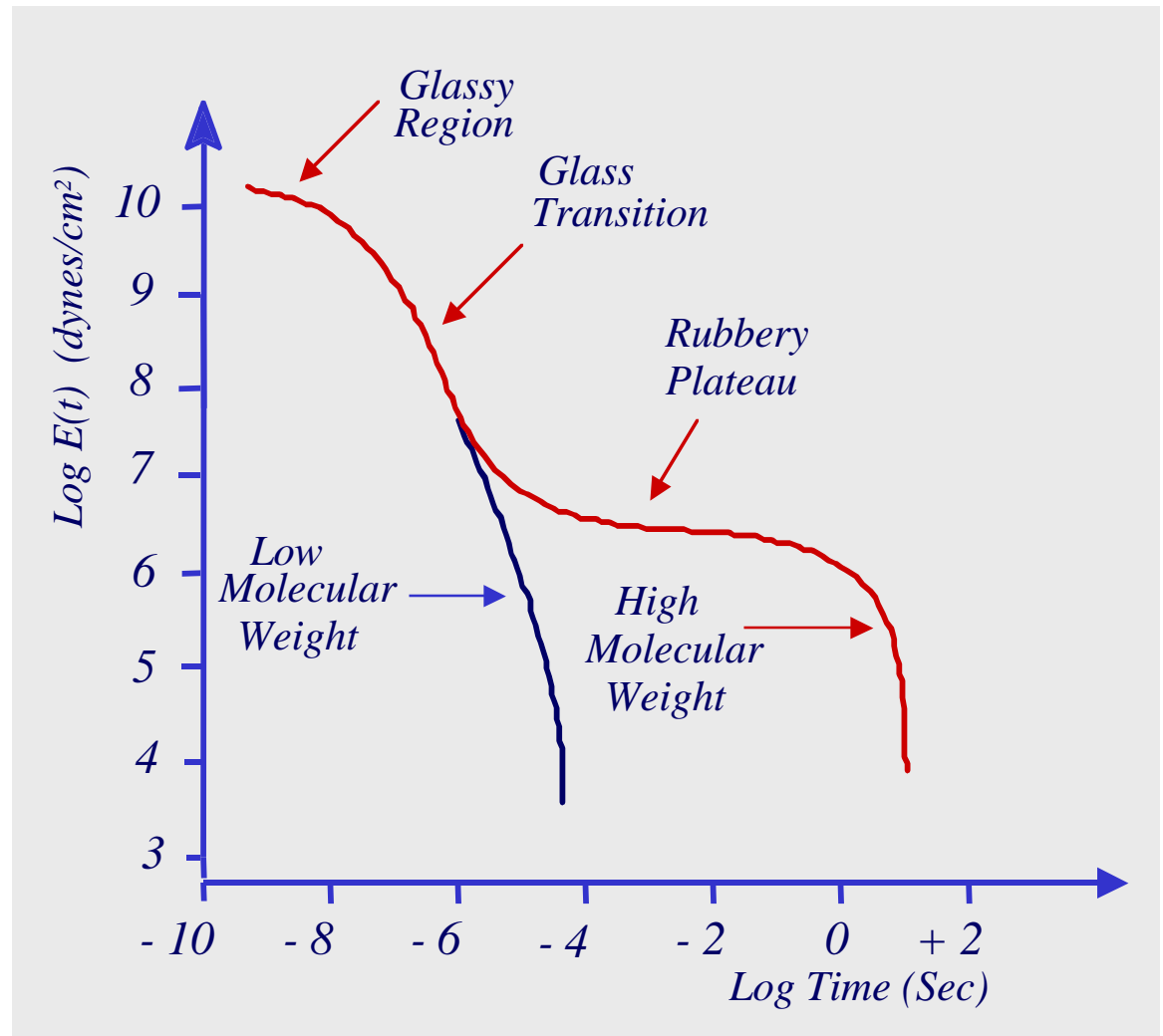


Measured over
Some arbitrary
Time period - say
10 secs

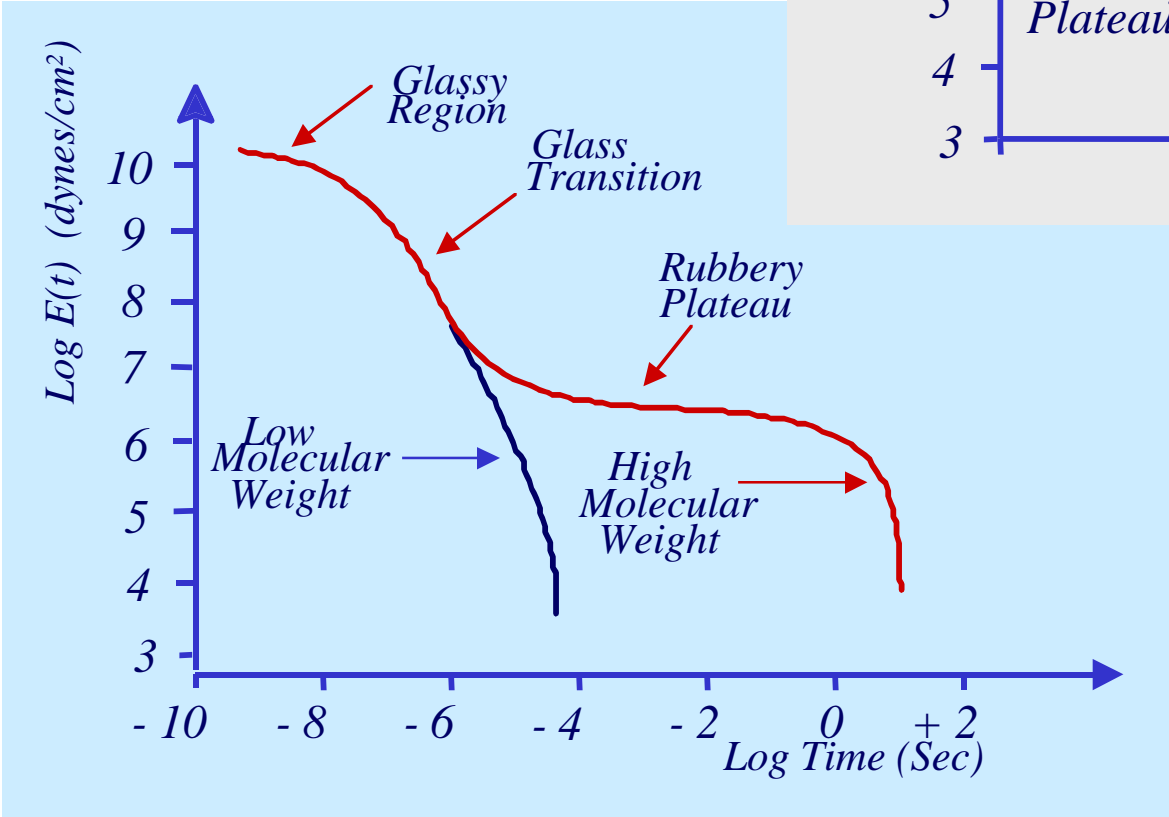
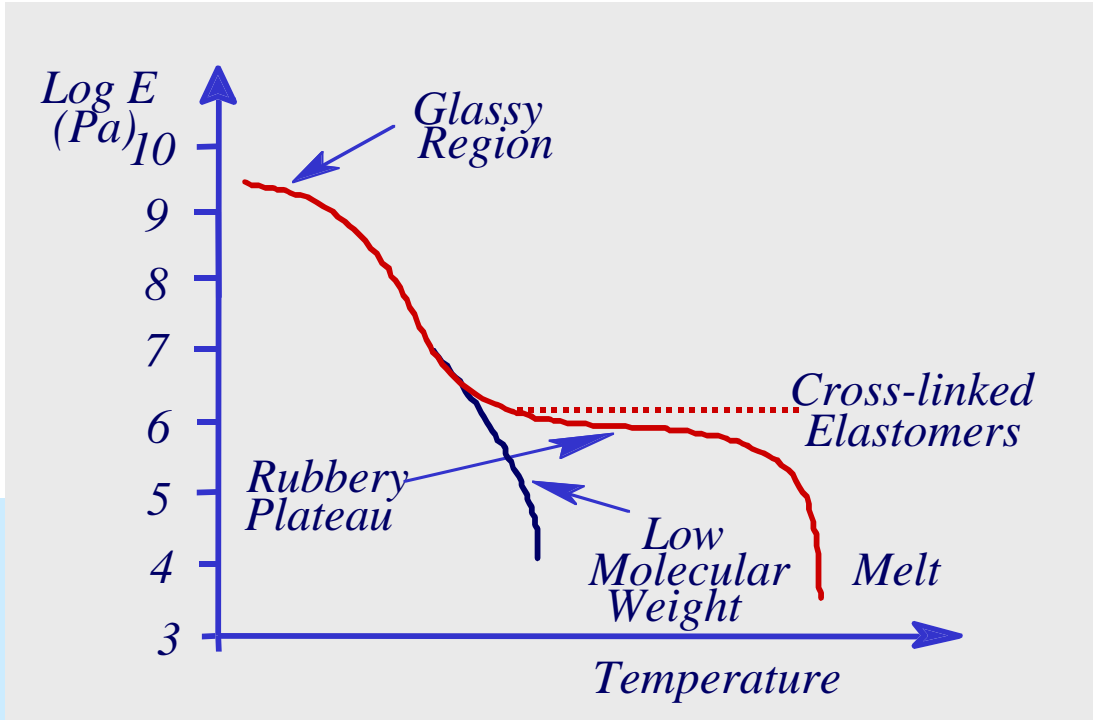
Viscoelastic Properties Of Amorphous Polymers

Stretch sample an arbitrary amount, measure the stress required to maintain this strain.

$$\text{Then } E(t) = \sigma(t)/\epsilon$$

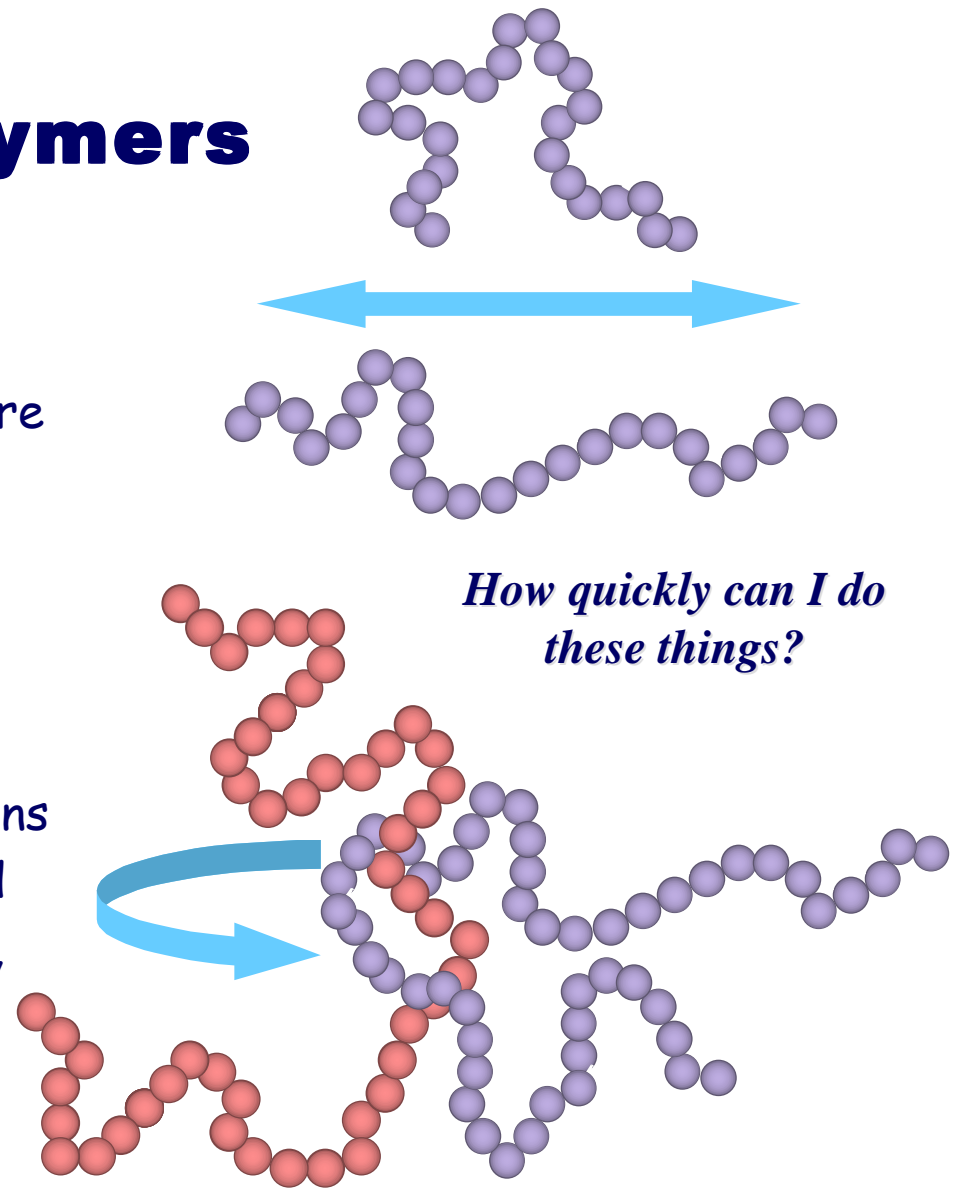


Time Temperature Equivalence

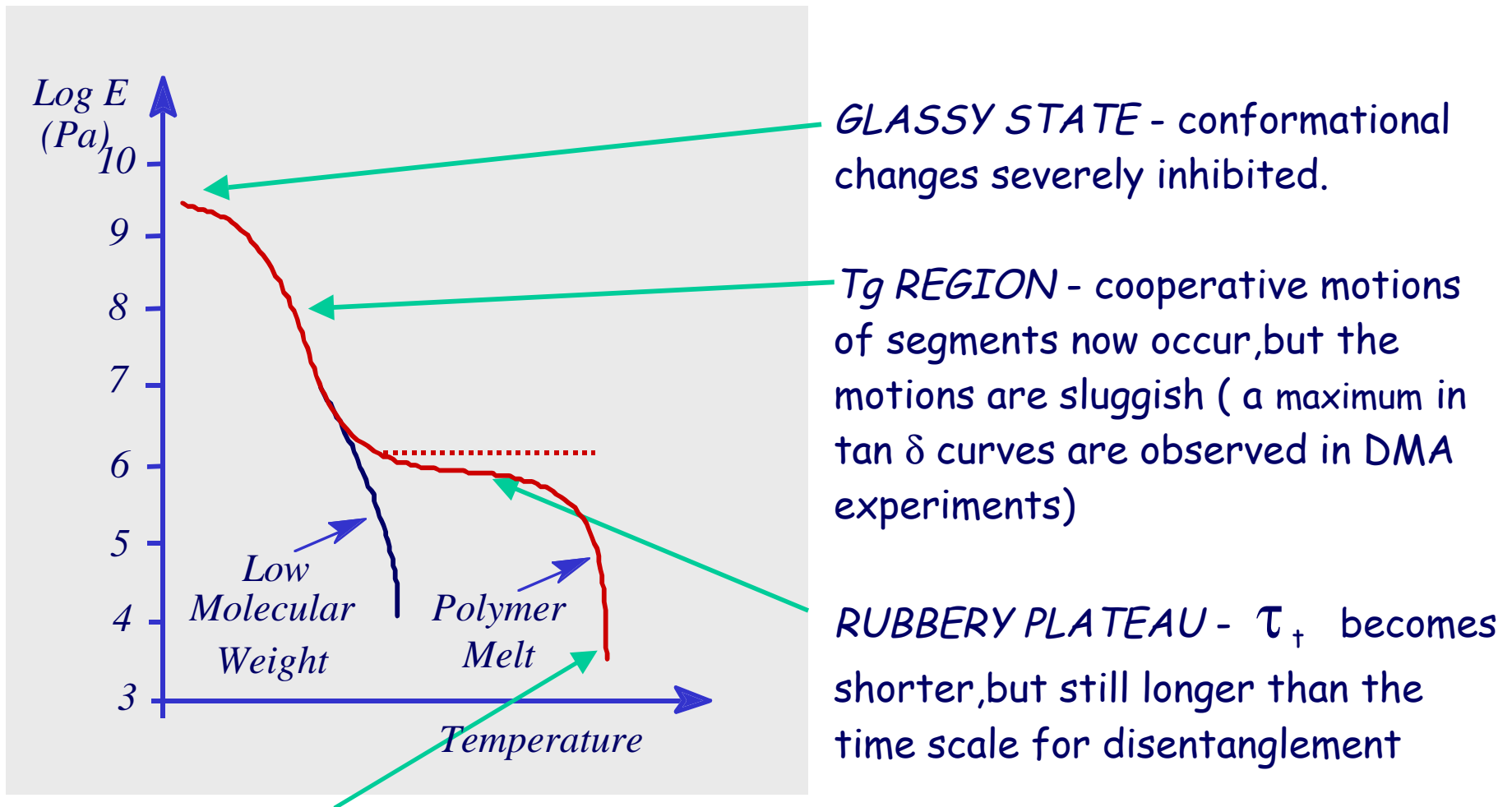


Relaxation in Polymers

First consider a hypothetical isolated chain in space, then imagine stretching this chain instantaneously so that there is a new end - to - end distance. The distribution of bond angles (trans, gauche, etc) changes to accommodate the conformations that are allowed by the new constraints on the ends. Because it takes time for bond rotations to occur, particularly when we also add in the viscous forces due to neighbors, we say the chain *RELAXES* to the new state and the relaxation is described by a characteristic time τ .

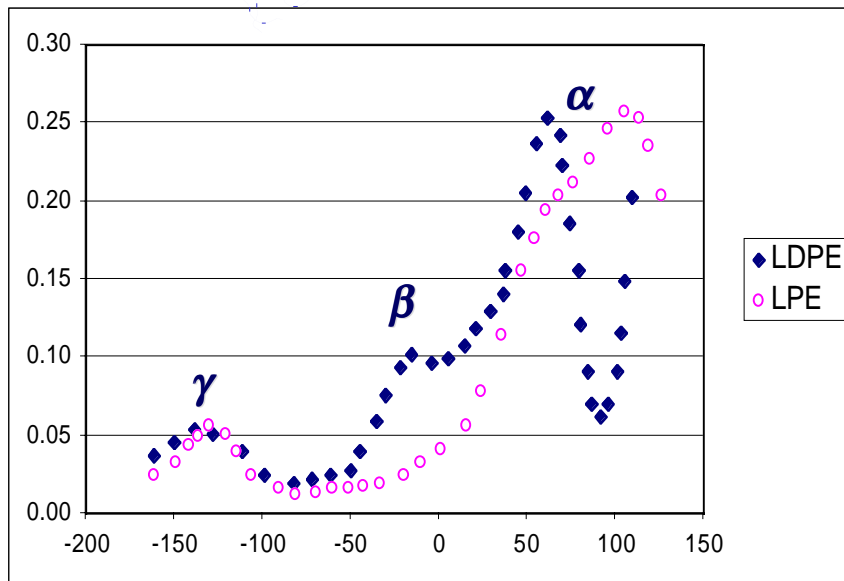
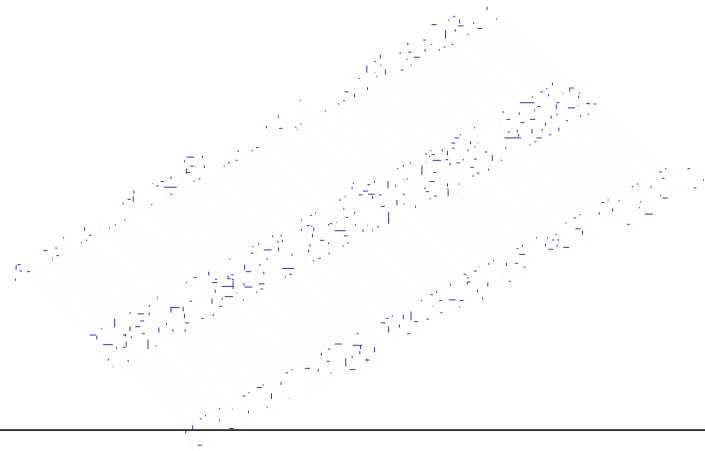


Amorphous Polymers - the Four Regions of Viscoelastic Behavior



TERMINAL FLOW - the time scale for disentanglement becomes shorter and the melt becomes more fluid like in its behavior

Semicrystalline Polymers



Redrawn from the data of H. A. Flocke,
Kolloid-Z. Z. Polym., **180**, 188 (1962).

- Motion in the amorphous domains constrained by crystallites
- Motions above T_g are often more complex, often involving coupled processes in the crystalline and amorphous domains
- Less easy to generalize - polymers often have to be considered individually - see DMA data opposite

Topics to be Covered

- *Simple models of Viscoelastic Behavior*
- *Time-Temperature Superposition Principle*

Chapter 11 in CD (Polymer Science and Engineering)

Mechanical and Theoretical Models of Viscoelastic Behavior

GOAL - relate $G(t)$ and $J(t)$ to relaxation behavior.

We will only consider *LINEAR MODELS*. (i.e. if we double $G(t)$ [or $\sigma(t)$], then $\gamma(t)$ [or $\epsilon(t)$] also increases by a factor of 2 (small loads and strains)).

TENSILE EXPERIMENT

SHEAR EXPERIMENT

Stress Relaxation

$$E(t) = \frac{\sigma(t)}{\epsilon_0}$$

$$G(t) = \frac{\tau(t)}{\gamma_0}$$

Creep

$$D(t) = \frac{\epsilon(t)}{\sigma_0}$$

$$J(t) = \frac{\gamma(t)}{\tau_0}$$

Linear Time Independent Behavior

$$E = \frac{1}{D}$$

$$G = \frac{1}{J}$$

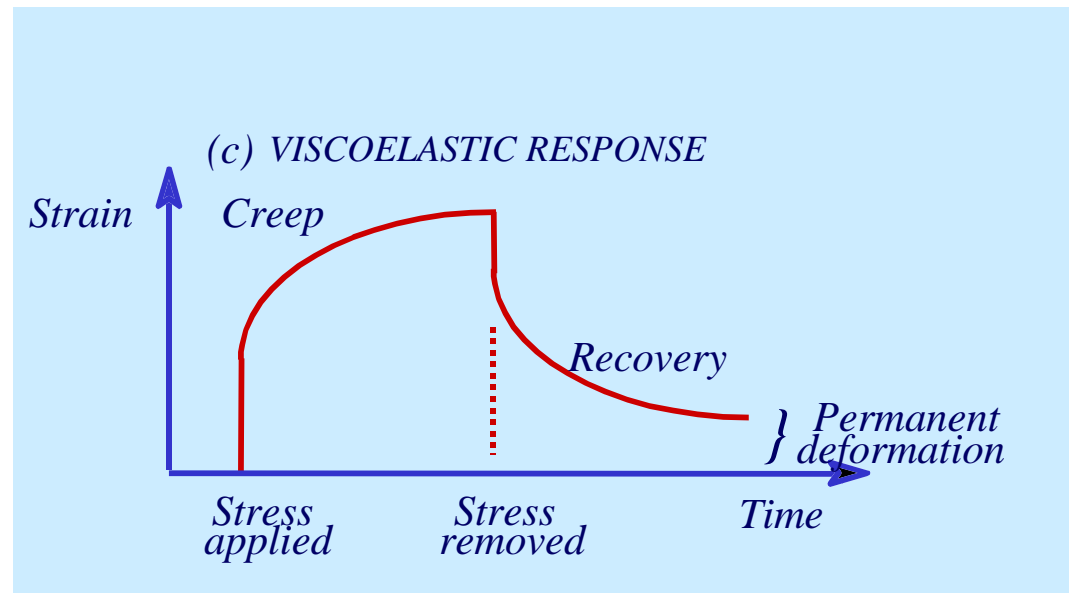
Time Dependent Behavior

$$E(t) \neq \frac{1}{D(t)}$$

$$G(t) \neq \frac{1}{J(t)}$$

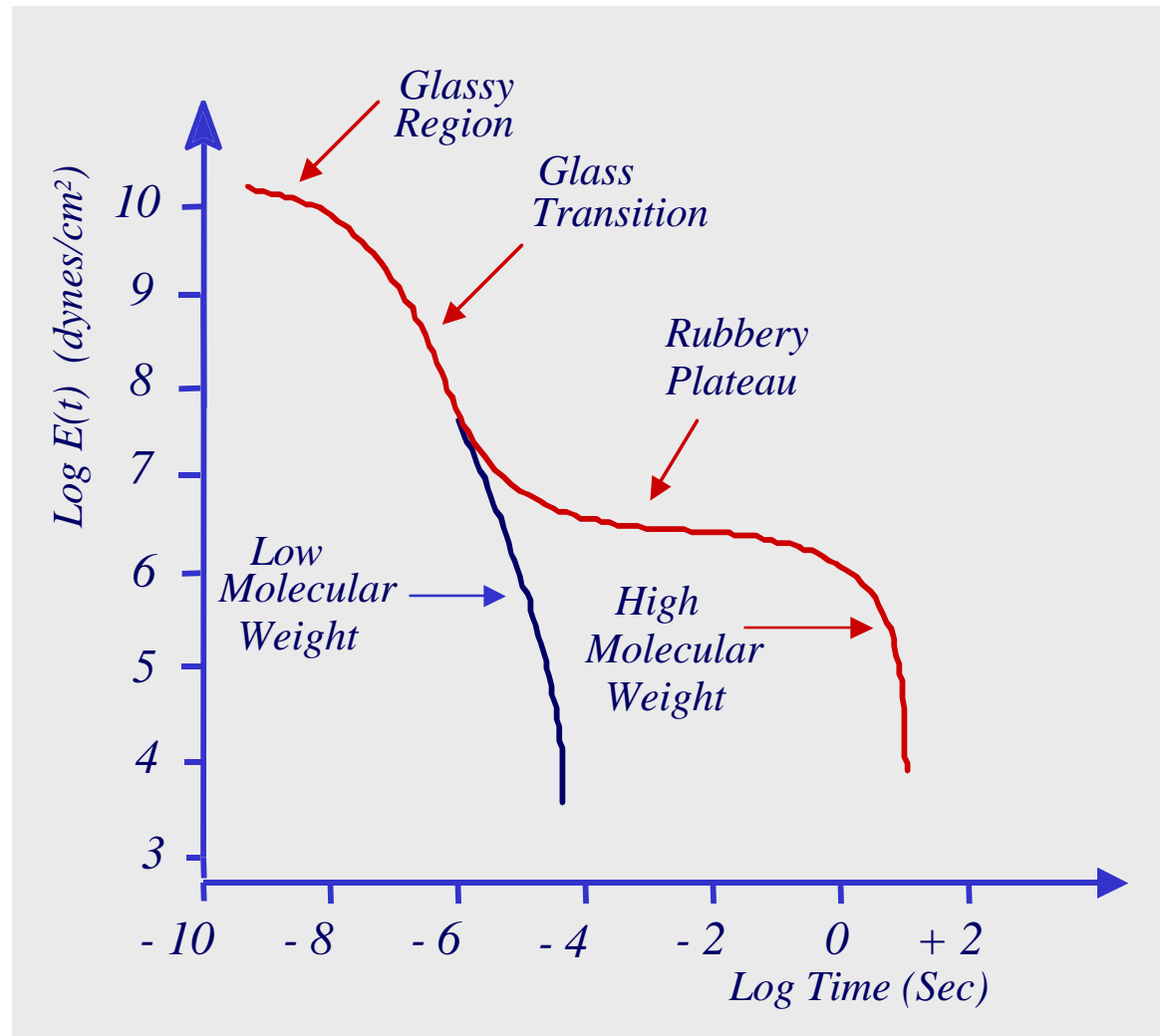
Simple Models of the Viscoelastic Behavior of Amorphous Polymers

Keep in mind that simple creep and recovery data for viscoelastic materials looks something like this

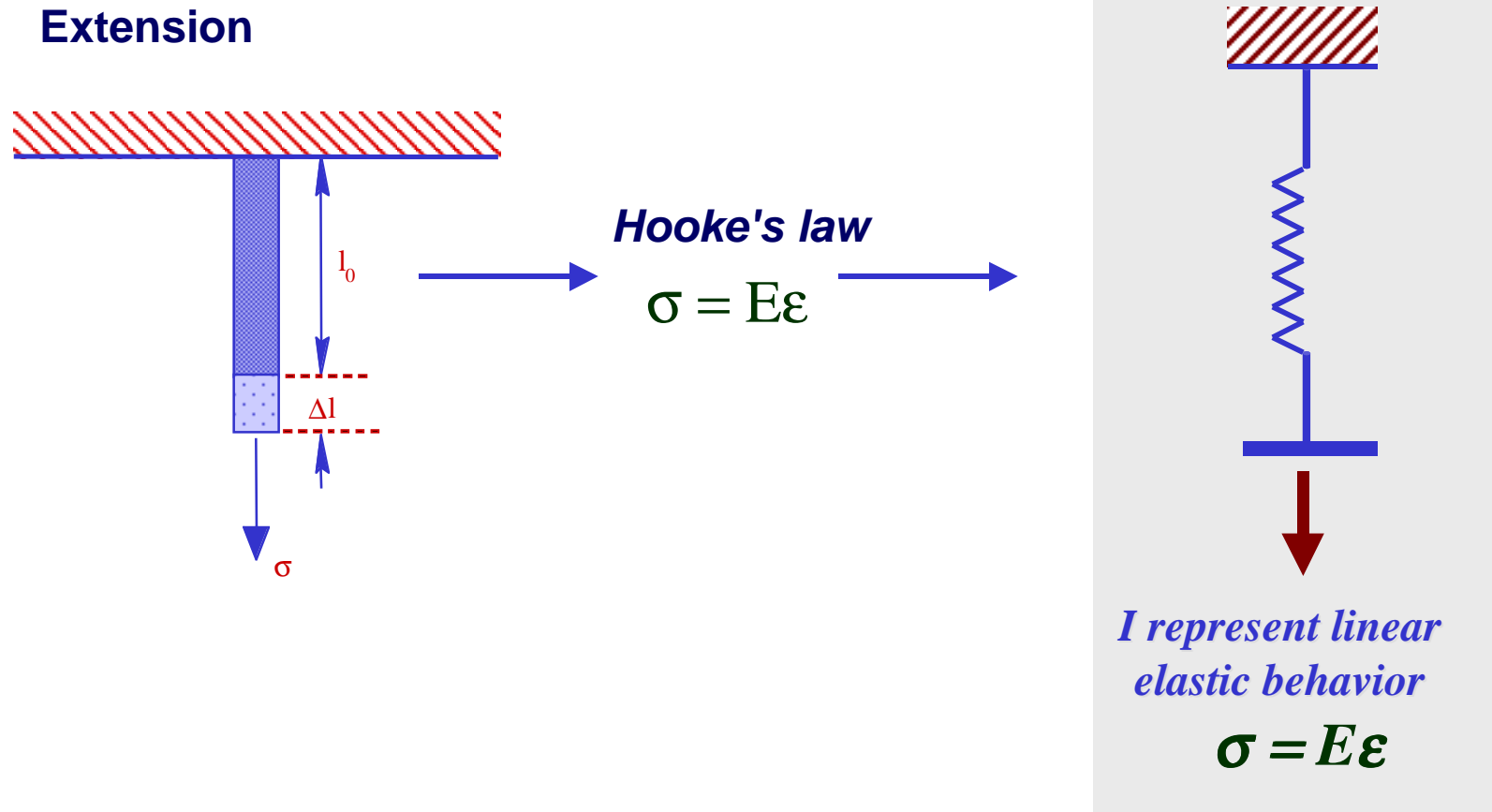


Simple Models of the Viscoelastic Behavior of Amorphous Polymers

While stress relaxation data look something like this;

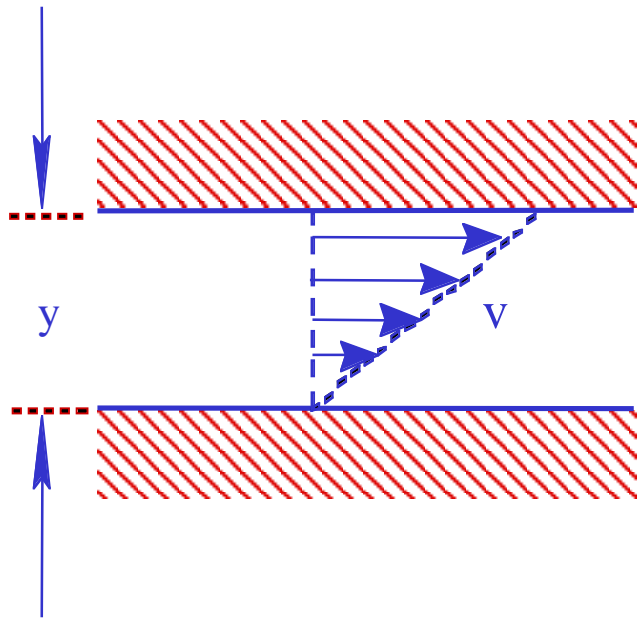


Simple Models of the Viscoelastic Behavior of Amorphous Polymers



Simple Models of the Viscoelastic Behavior of Amorphous Polymers

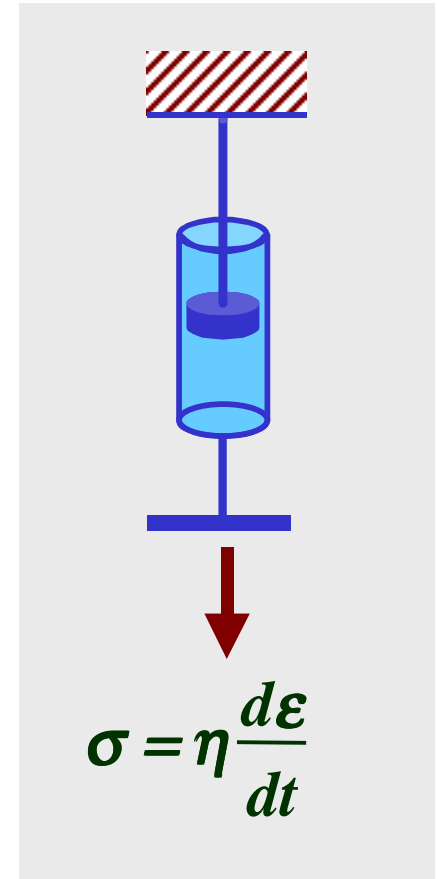
Viscous flow



v_0

Newtonian fluid

$$\tau_{xy} = \eta \dot{\gamma}$$

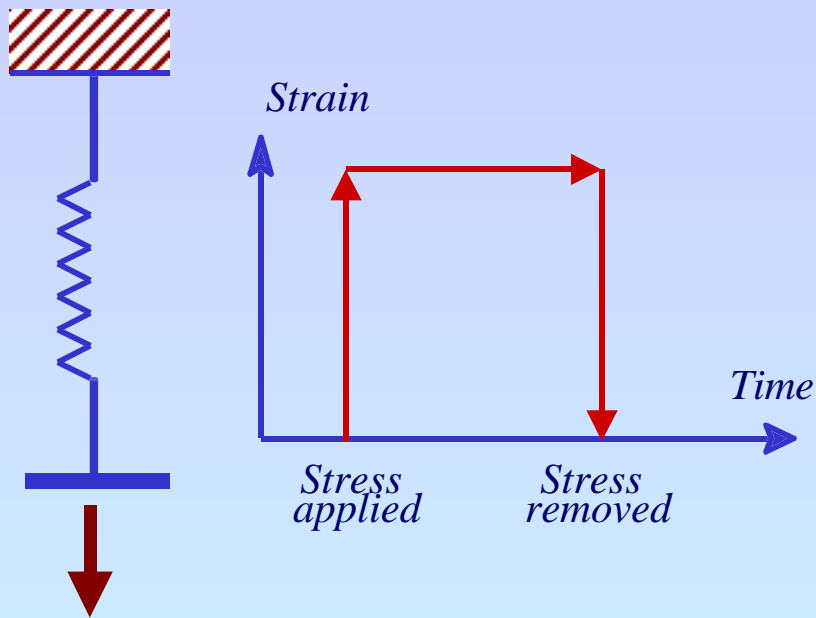


$$\sigma = \eta \frac{d\varepsilon}{dt}$$

Strain vs. Time for Simple Models

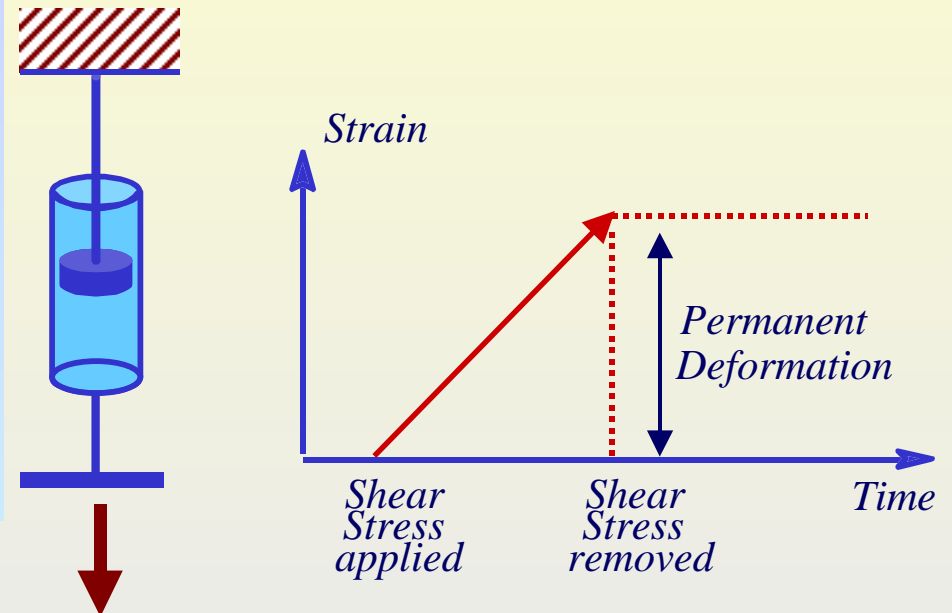
PURELY ELASTIC RESPONSE

$$\sigma = E\varepsilon$$



PURELY VISCOUS RESPONSE

$$\sigma = \eta \frac{d\varepsilon}{dt}$$



Maxwell Model

Maxwell was interested in creep and stress relaxation and developed a differential equation to describe these properties

Maxwell started with Hooke's law

$$\sigma = E\varepsilon$$

Then allowed σ to vary with time

$$\frac{d\sigma}{dt} = E \frac{d\varepsilon}{dt}$$

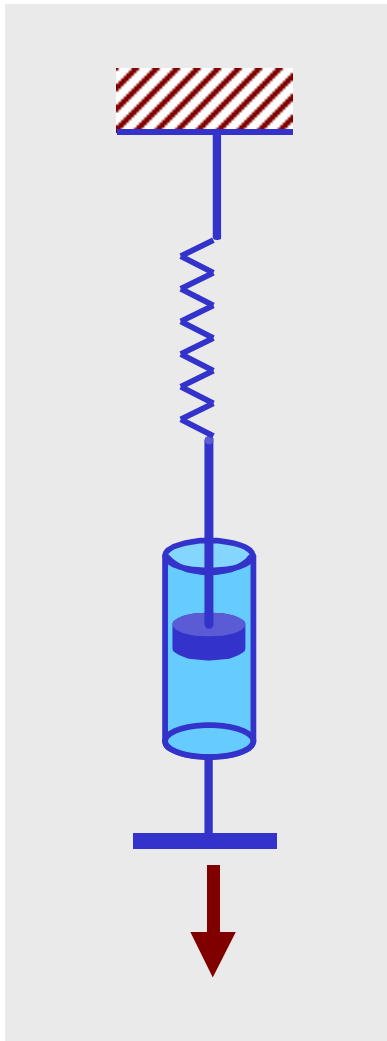
Writing for a Newtonian fluid

$$\sigma = \eta \frac{d\varepsilon}{dt}$$

Then assuming that the rate of strain is simply a sum of these two contributions

$$\frac{d\varepsilon}{dt} = \frac{\sigma}{\eta} + \frac{1}{E} \frac{d\sigma}{dt}$$

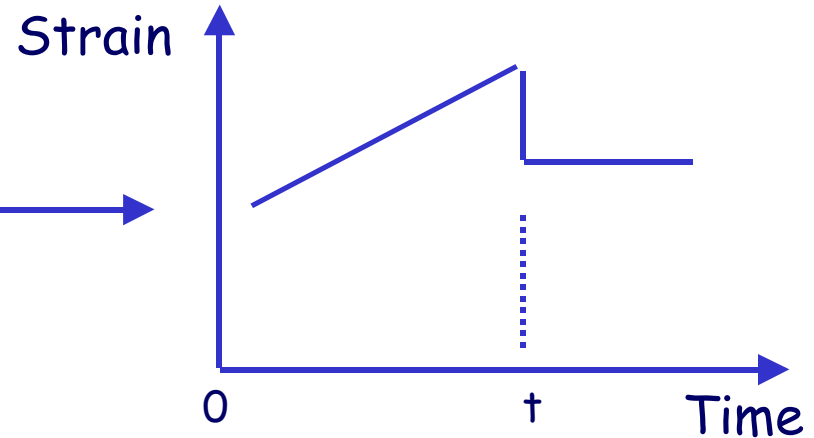
MAXWELL MODEL - *Creep and Recovery*



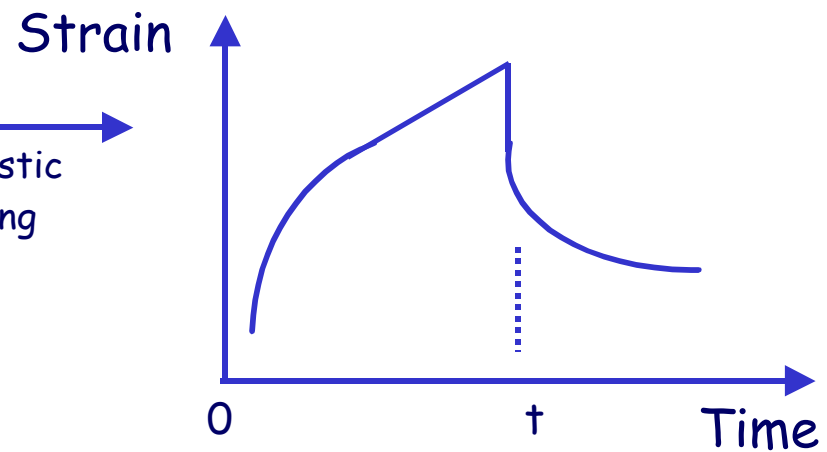
A picture representation of Maxwell's equation

Creep and recovery

$$\frac{d\varepsilon}{dt} = \frac{\sigma}{\eta}$$



Recall that real viscoelastic behaviour looks something like this



Maxwell Model -Stress Relaxation

$$\frac{d\varepsilon}{dt} = \frac{\sigma}{\eta} + \frac{1}{E} \frac{d\sigma}{dt}$$

In a stress relaxation experiment

$$\frac{d\varepsilon}{dt} = 0$$

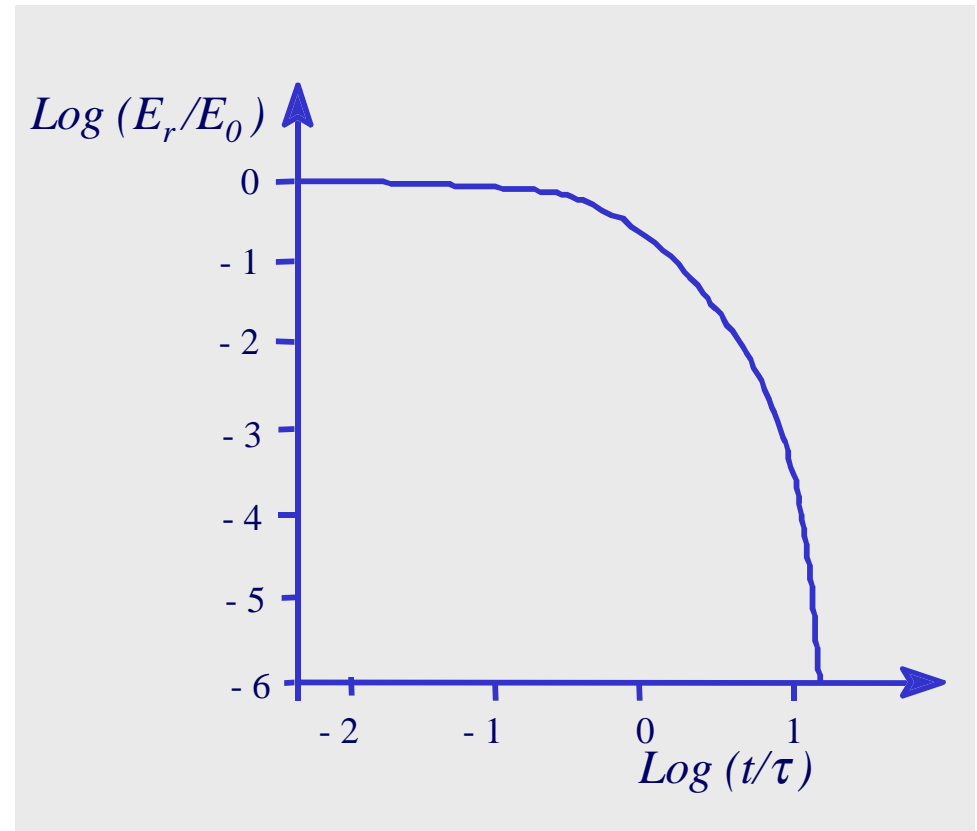
Hence
$$\frac{d\sigma}{\sigma} = -\frac{1}{\tau} dt$$

$$\sigma = \sigma_0 \exp[-t/\tau]$$

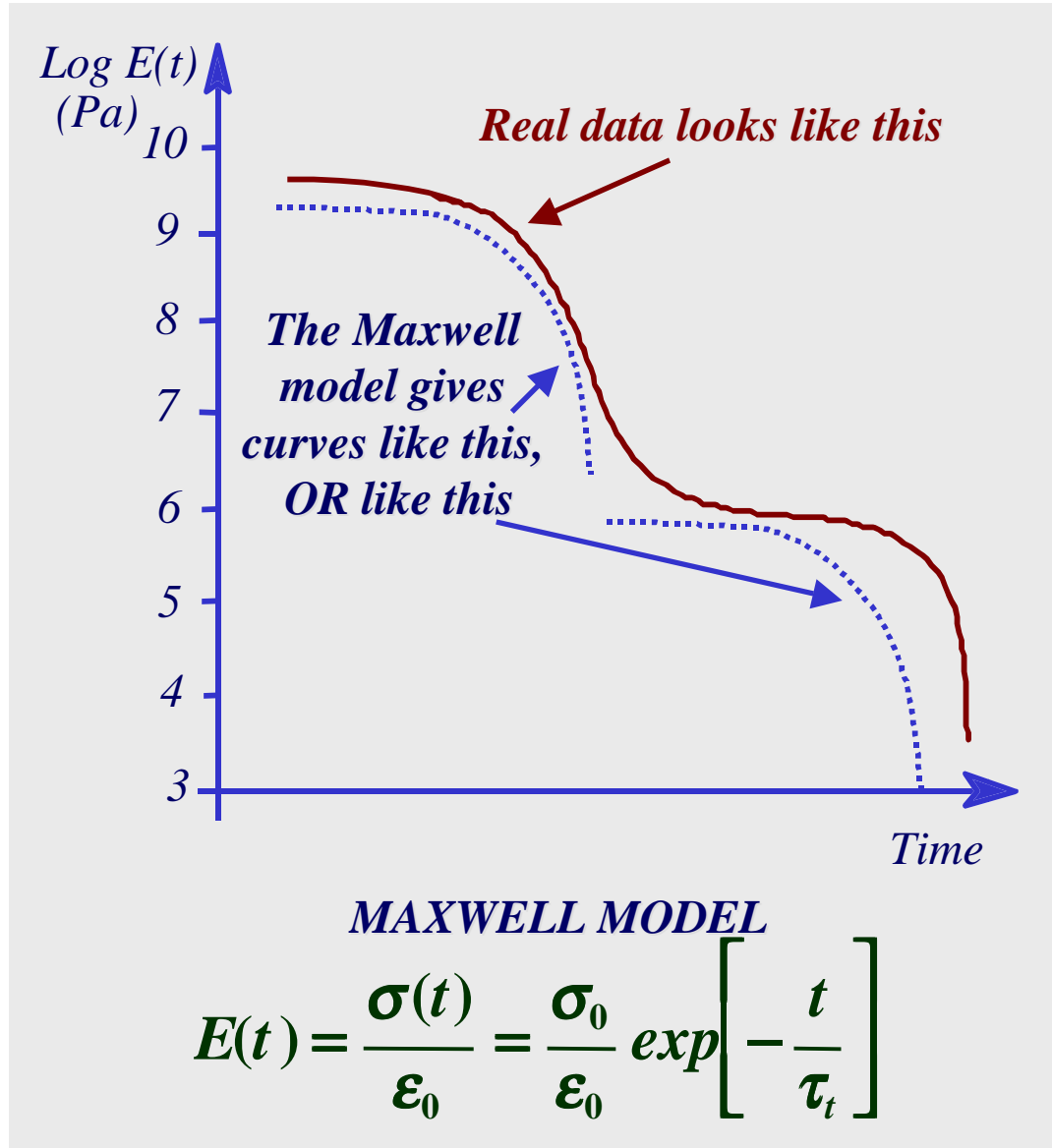
Where

$$\tau = \frac{\eta}{E}$$

Relaxation time



Maxwell Model - Stress Relaxation



Voigt Model

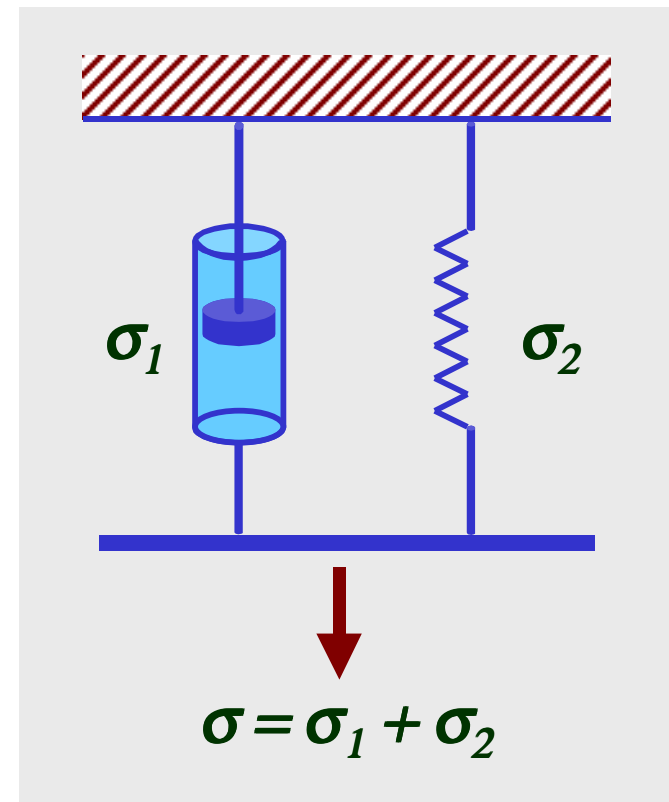
Maxwell model essentially assumes a uniform distribution of stress. Now assume uniform distribution of strain -
VOIGT MODEL

Picture representation →

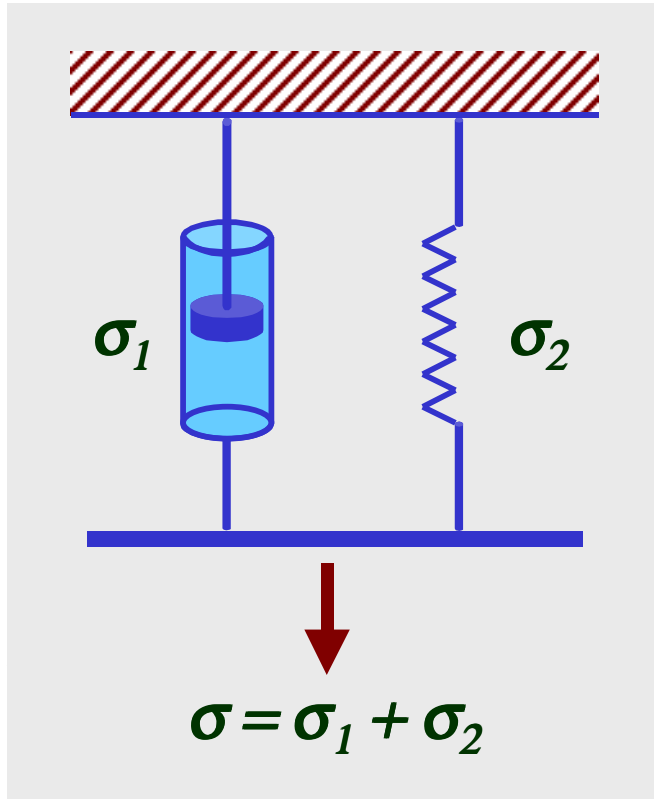
Equation

$$\sigma(t) = E\varepsilon(t) + \eta \frac{d\varepsilon(t)}{dt}$$

(Strain in both elements of the model is the same and the total stress is the sum of the two contributions)



Voigt Model - Creep and Stress Relaxation



Gives a retarded elastic response but does not allow for "ideal" stress relaxation, in that the model cannot be "instantaneously" deformed to a given strain.

But in *CREEP* $\sigma = \text{constant}, \sigma_0$

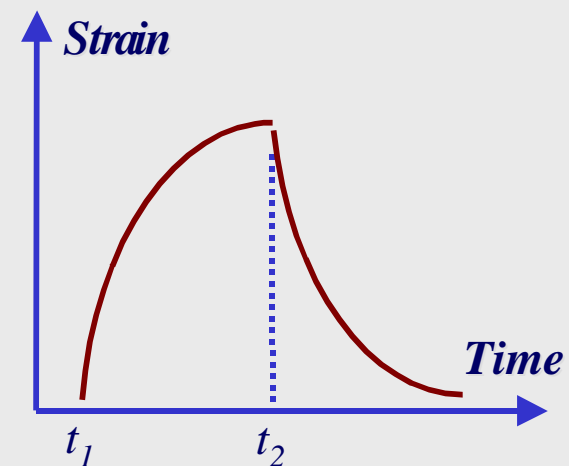
$$\sigma(t) = \sigma_0 = E\varepsilon(t) + \eta \frac{d\varepsilon(t)}{dt}$$

$$\frac{d\varepsilon(t)}{dt} + \frac{\varepsilon(t)}{\tau_t'} = \frac{\sigma_0}{\eta}$$

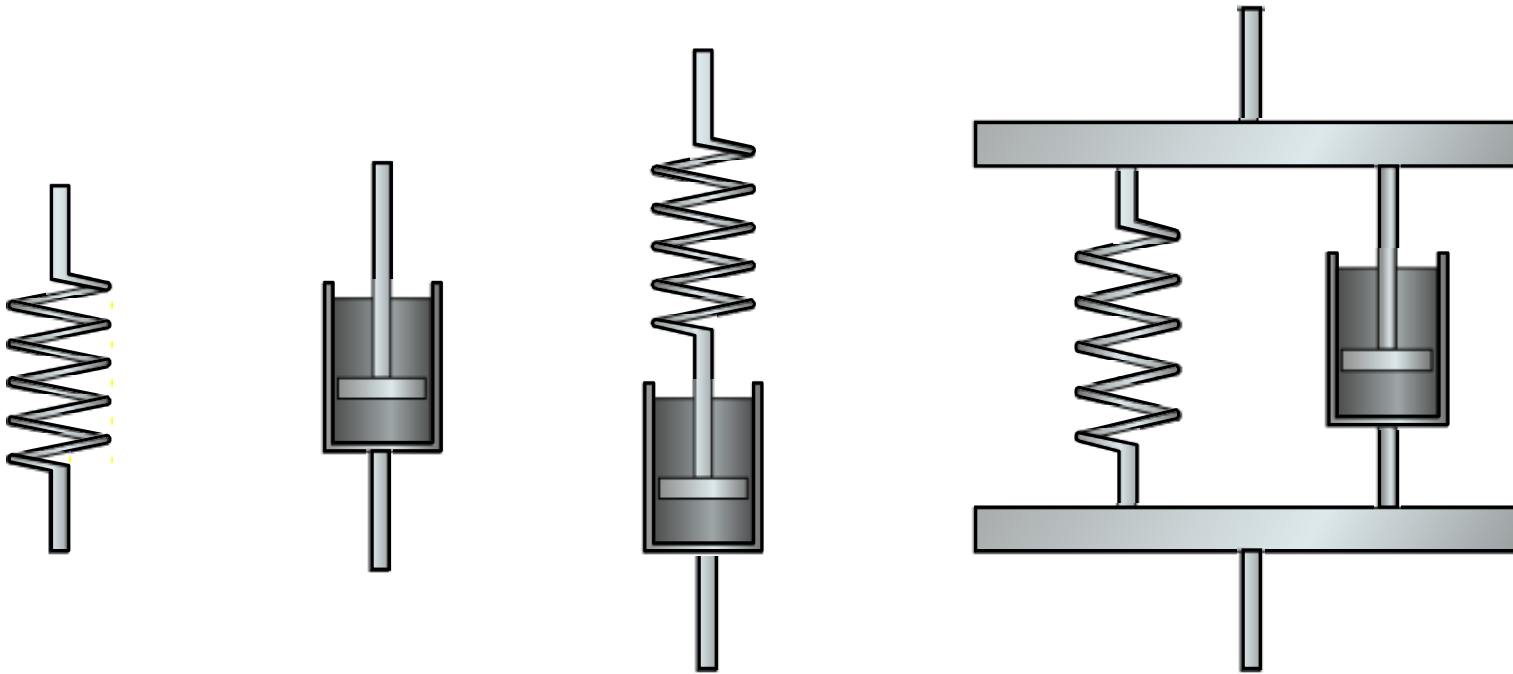
$$\varepsilon(t) = \frac{\sigma_0}{E} [1 - \exp(-t/\tau_t')]$$

τ_t' - retardation time (η/E)

RETARDED ELASTIC RESPONSE

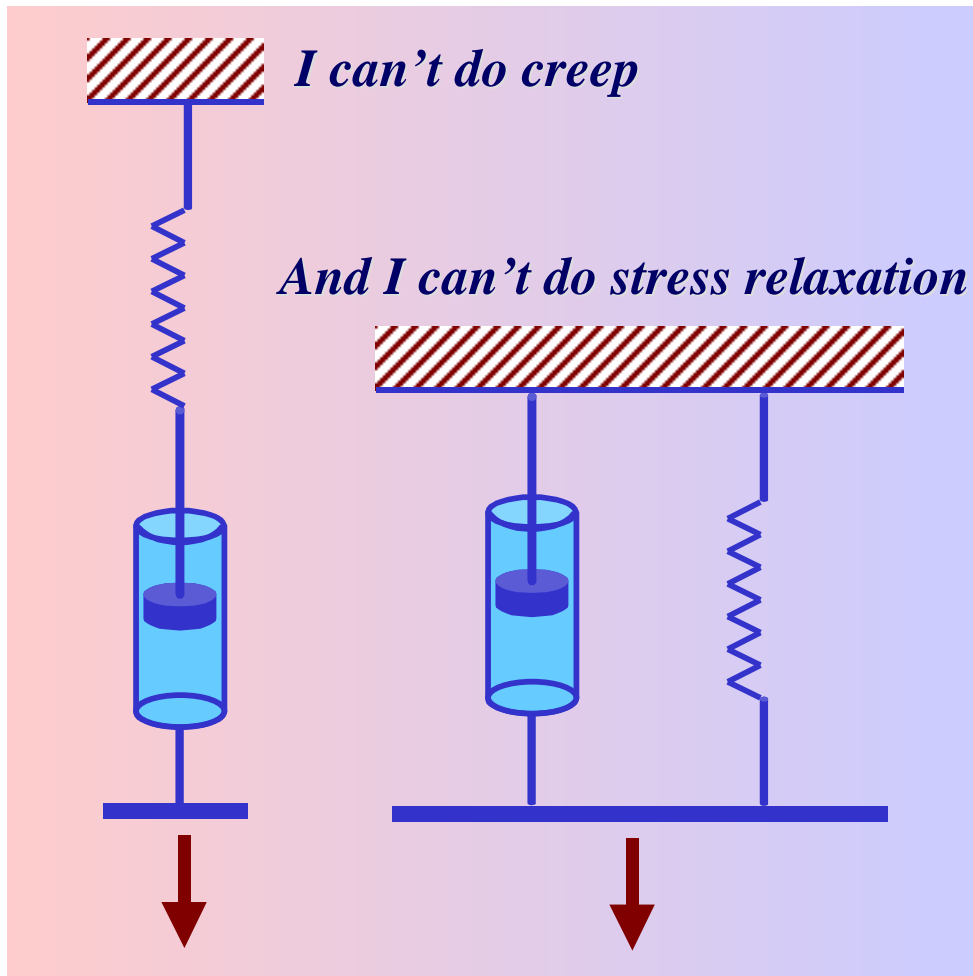


Summary



What do the strain/time plots look like?

Problems with Simple Models



- The Maxwell model cannot account for a retarded elastic response
- The Voigt model does not describe stress relaxation
- Both models are characterized by single relaxation times - a spectrum of relaxation times would provide a better description

NEXT - CONSIDER THE FIRST TWO PROBLEMS

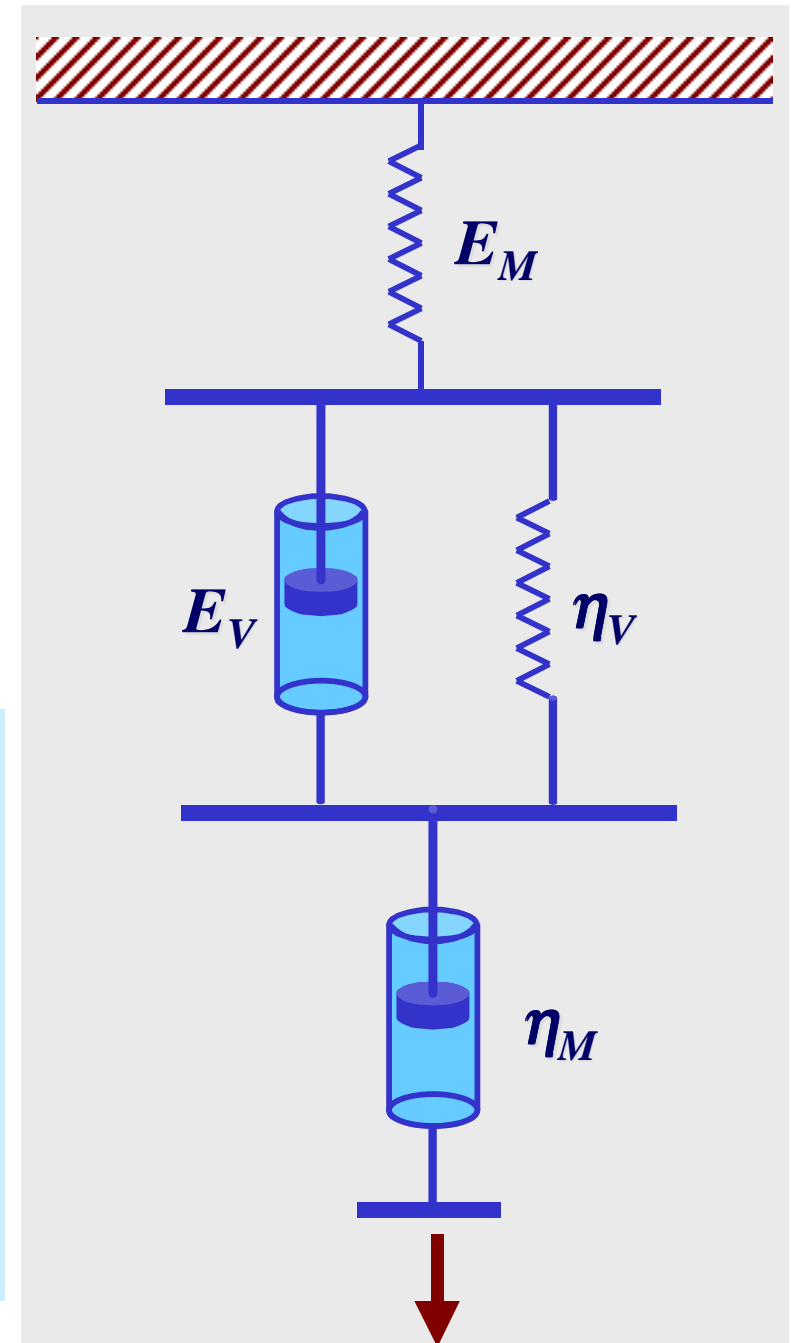
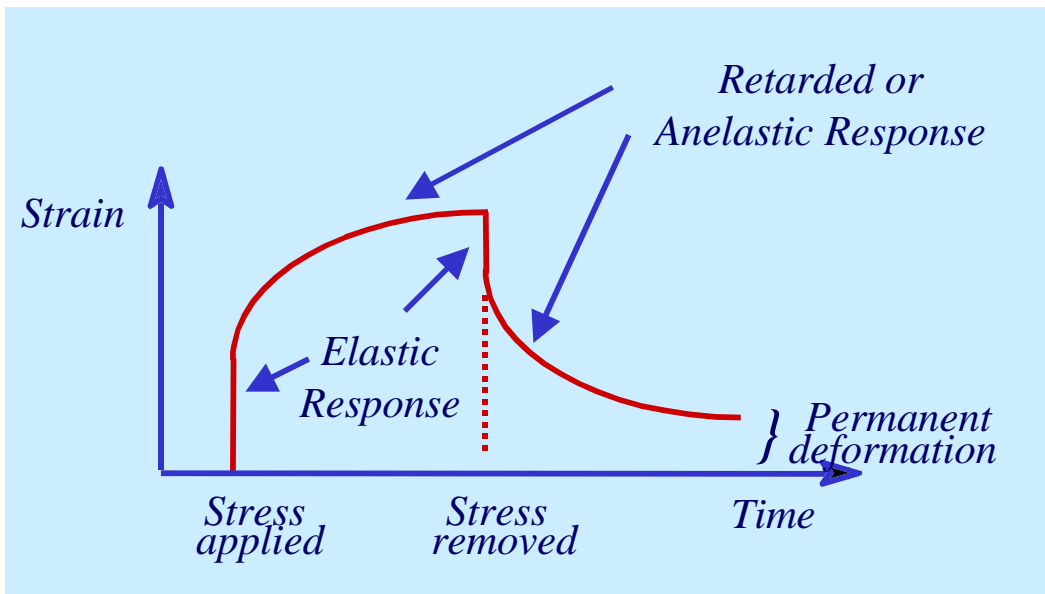
THEN - THE PROBLEM OF A SPECTRUM OF RELAXATION TIMES

Four - Parameter Model

Elastic + viscous flow + retarded elastic

Eg CREEP

$$\varepsilon = \frac{\sigma_0}{E_M} + \frac{\sigma_0 t}{\eta_M} + \frac{\sigma_0}{E_M} [1 - \exp(-t/\tau_t)]$$



Distributions of Relaxation and Retardation Times

We have mentioned that although the Maxwell and Voigt models are seriously flawed, the equations have the right form.

What we mean by that is shown opposite, where equations describing the Maxwell model for stress relaxation and the Voigt model for creep are compared to equations that account for a continuous range of relaxation times.

These equations can be obtained by assuming that relaxation occurs at a rate that is linearly proportional to the distance from equilibrium and use of the Boltzmann superposition principle. We will show how the same equations are obtained from models.

Stress Relaxation

$$E(t) = E_0 \exp(-t / \tau_t)$$

$$E(t) = \int_0^{\infty} E(\tau_t) \exp(-t / \tau_t) d\tau_t$$

Creep

$$D(t) = D_0 [1 - \exp(-t / \tau_t')]]$$

$$D(t) = \int_0^{\infty} D(\tau_t') [1 - \exp(-t / \tau_t')] d\tau_t'$$

The Maxwell - Wiechert Model

$$\begin{aligned}\frac{d\varepsilon}{dt} &= \frac{\sigma_1}{\eta_1} + \frac{1}{E_1} \frac{d\sigma_1}{dt} \\ &= \frac{\sigma_2}{\eta_2} + \frac{1}{E_2} \frac{d\sigma_2}{dt} \\ &= \frac{\sigma_3}{\eta_3} + \frac{1}{E_3} \frac{d\sigma_3}{dt}\end{aligned}$$

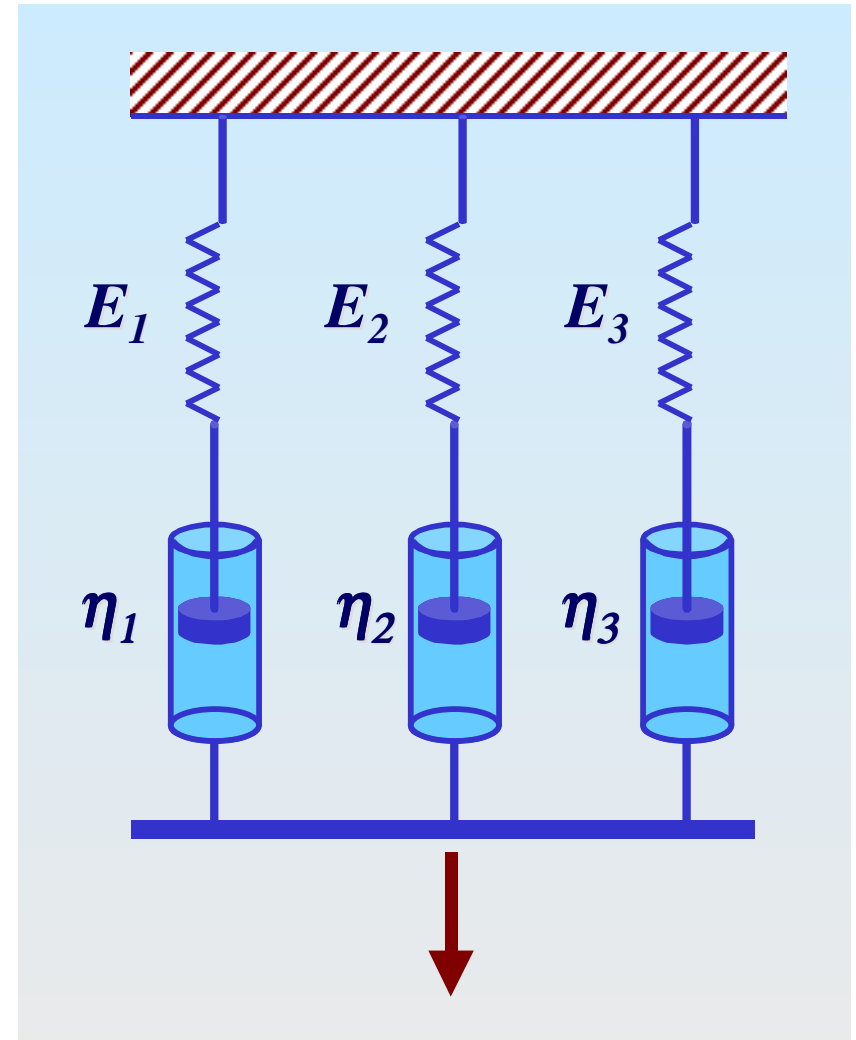
Consider stress relaxation

$$\frac{d\varepsilon}{dt} = 0$$

$$\sigma_1 = \sigma_0 \exp[-t/\tau_{t1}]$$

$$\sigma_2 = \sigma_0 \exp[-t/\tau_{t2}]$$

$$\sigma_3 = \sigma_0 \exp[-t/\tau_{t3}]$$

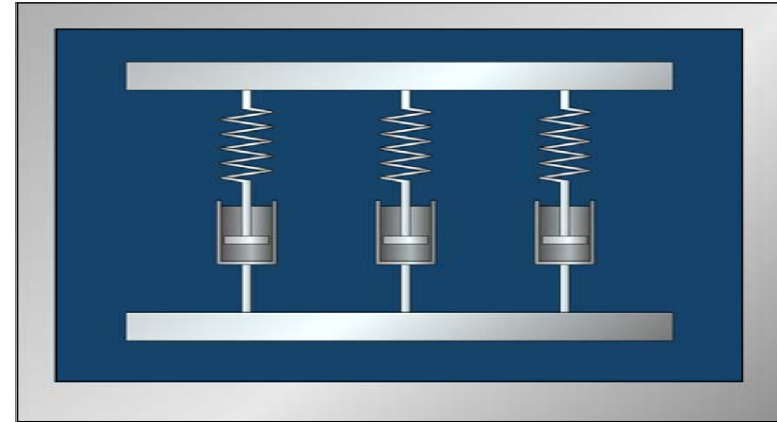


Distributions of Relaxation and Retardation Times

Stress relaxation modulus

$$E(t) = \sigma(t)/\varepsilon_0$$

$$\sigma(t) = \sigma_1 + \sigma_2 + \sigma_3$$



$$E(t) = \frac{\sigma_{01}}{\varepsilon_0} \exp(-t/\tau_{01}) + \frac{\sigma_{02}}{\varepsilon_0} \exp(-t/\tau_{02}) + \frac{\sigma_{03}}{\varepsilon_0} \exp(-t/\tau_{03})$$

Or, in general

$$E(t) = \sum E_n \exp(-t/\tau_{tn}) \quad \text{where} \quad E_n = \frac{\sigma_{0n}}{\varepsilon_0}$$

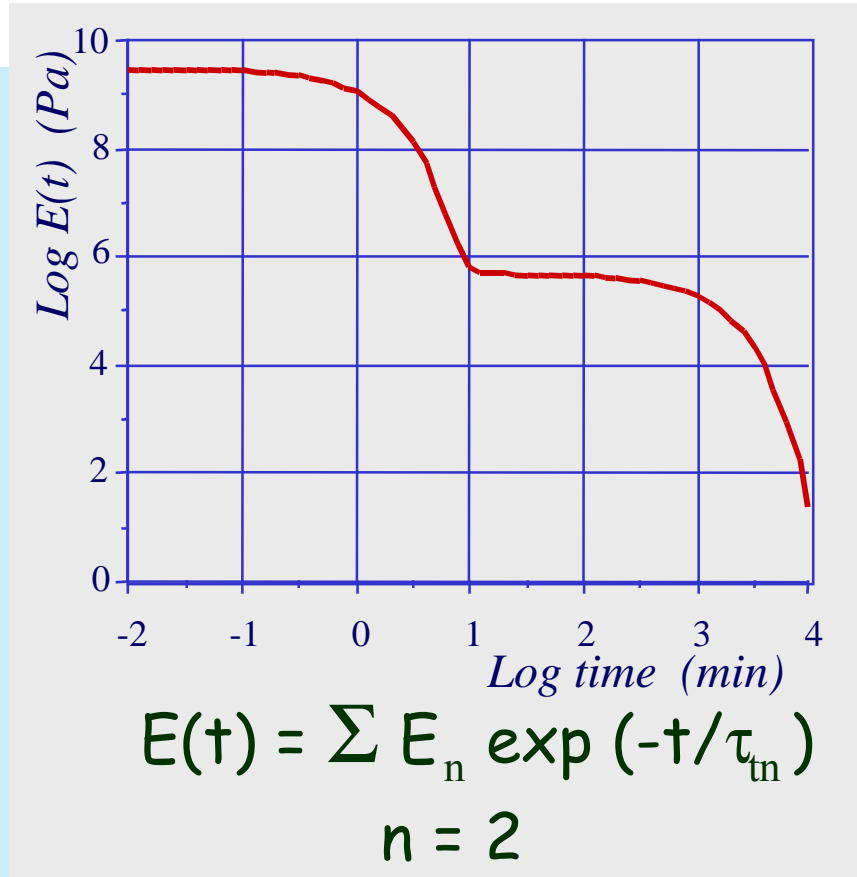
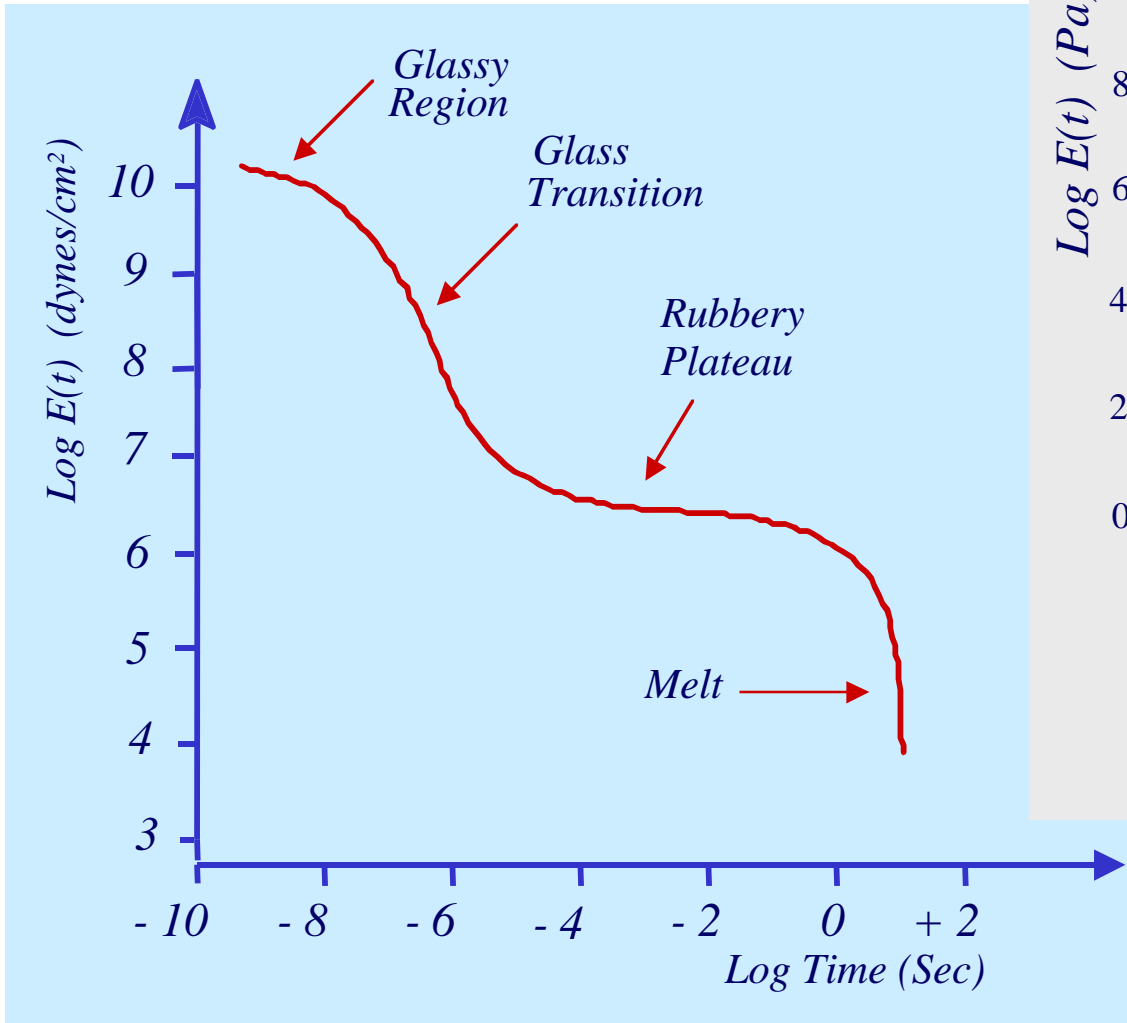
Similarly, for creep compliance combine voigt elements to obtain

$$D(t) = \sum D_n [1 - \exp(-t/\tau'_{tn})]$$

Distributions of Relaxation and Retardation Times

Example - the Maxwell - Wiechert Model With

$n = 2$

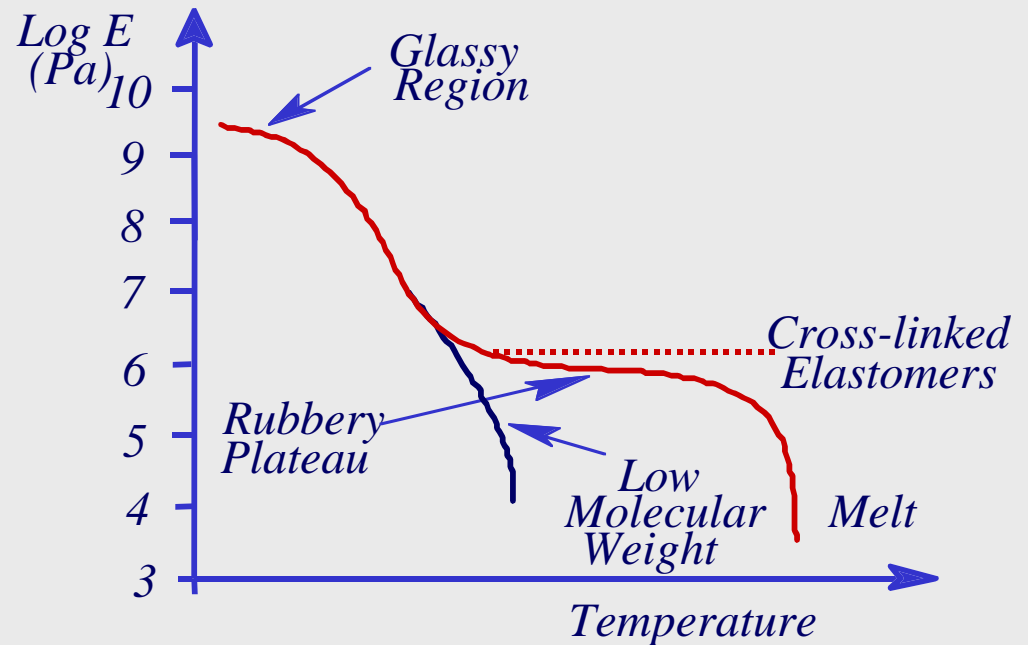
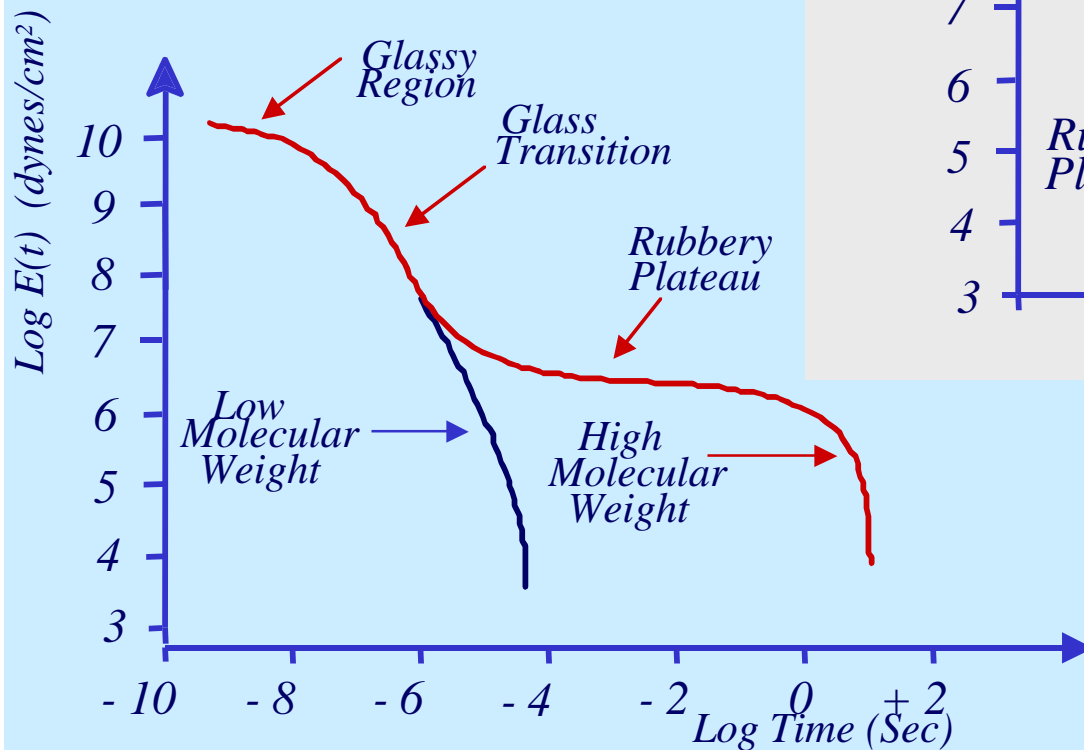


$$E(t) = \sum_{n=1}^2 E_n \exp(-t/\tau_{tn})$$

$n = 2$

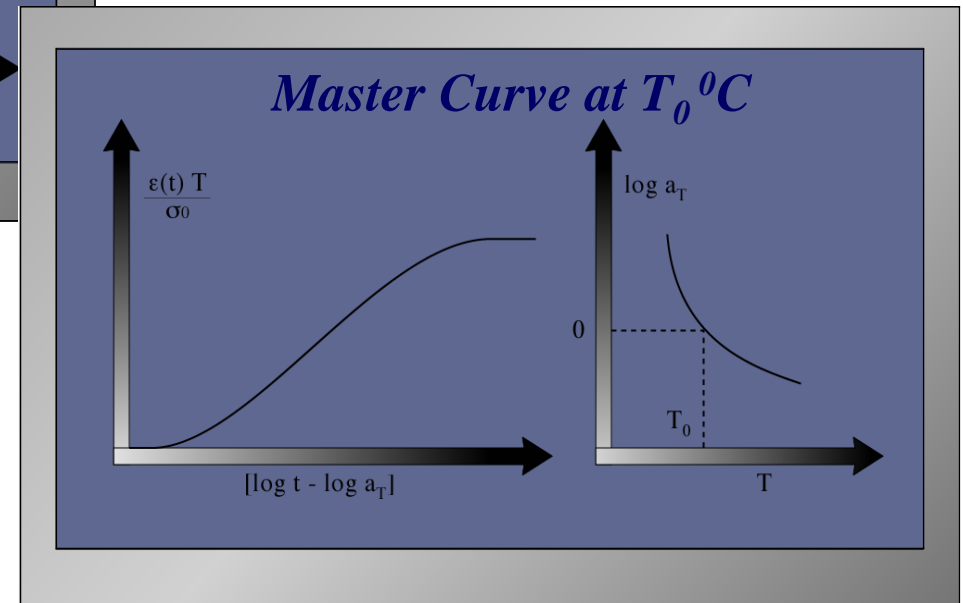
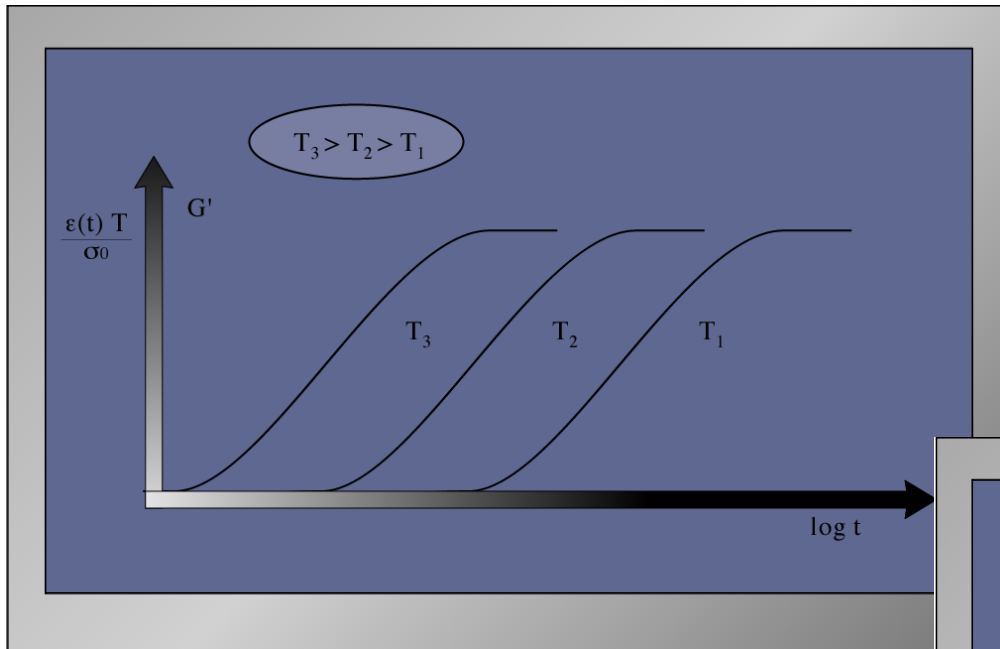
Time - Temperature Superposition Principle

Recall that we have seen that there is a time - temperature equivalence in behavior

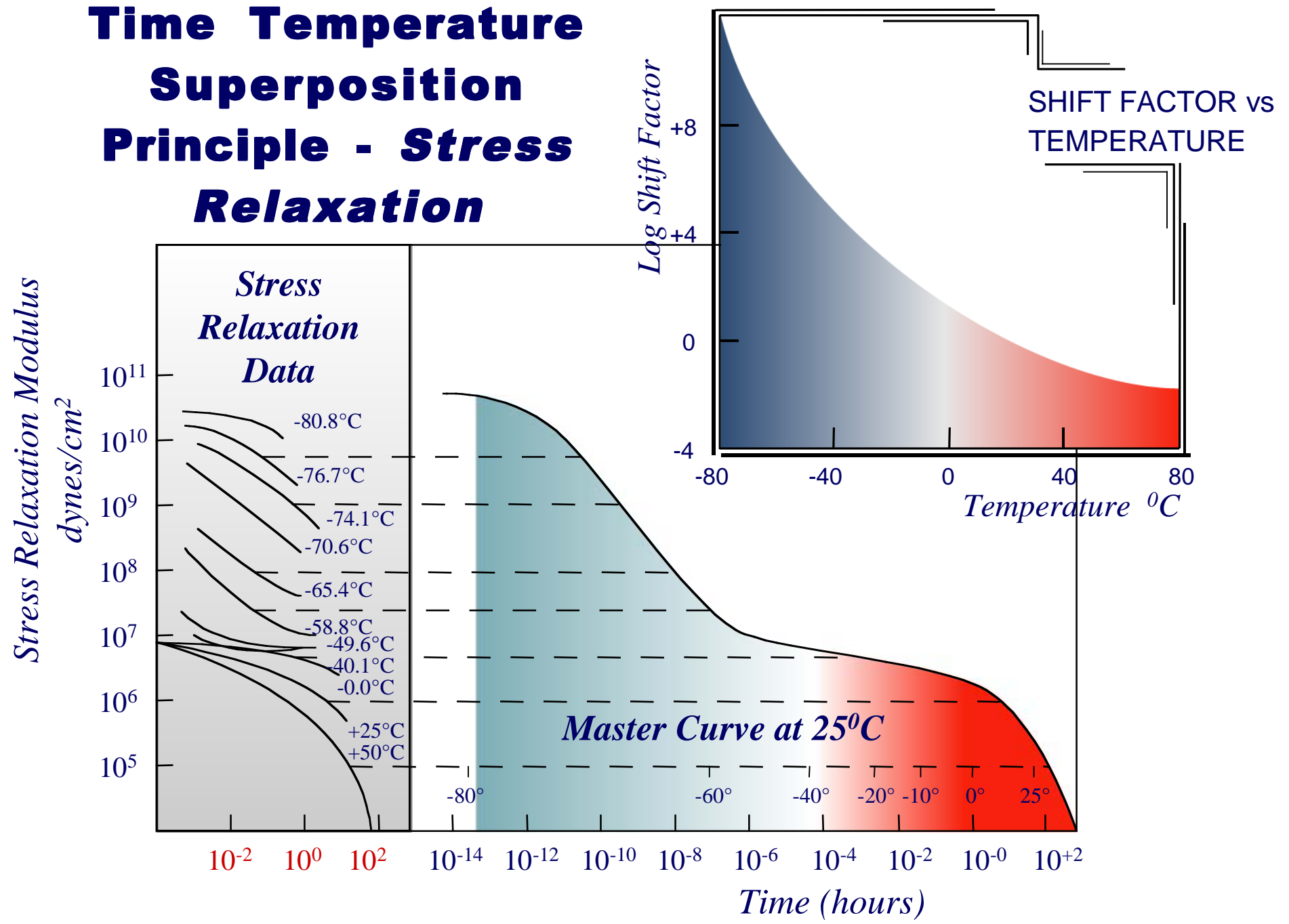


This can be expressed formally in terms of a superposition principle

Time Temperature Superposition Principle - Creep



Time Temperature Superposition Principle - Stress Relaxation



Relaxation Processes above T_g - the WLF Equation

From empirical observation

$$\text{Log } a_T = \frac{-C_1 (T - T_s)}{C_2 + (T - T_s)} \quad \text{For } T_g > T < \sim(T_g + 100^\circ\text{C})$$

Originally thought that C_1 and C_2 were universal constants, = 17.44 and 51.6, respectively, when $T_s = T_g$. Now known that these vary from polymer to polymer.

Homework problem - show how the WLF equation can be obtained from the relationship of viscosity to free volume as expressed in the Doolittle equation

Semi - Crystalline Polymers

NON - LINEAR RESPONSE TO STRESS. SIMPLE MODELS AND THE TIME - TEMPERATURE SUPERPOSITION PRINCIPLE DO NOT APPLY

